

# CHEMISTRY SEMESTER B: FINAL EXAM REVIEW EXAM QUESTIONS WITH 100% CORRET ANSWERS ALREADY GRADED A+

What atoms follow the Duet rule? - answer Li and Na

What type of bonding takes place between N and F? - answer covalent

How many electrons is an octet? - answer 8

To draw the Lewis Structure of the polyatomic ion  $\text{ClO}_2^-$ , you would have to \_\_\_\_\_ those in the structures of Cl, O, O and O. - answer add one electron to

The correct Lewis structure for a molecule of the compound  $\text{C}_2\text{H}_2$  contains - answer one triple bond

Which of the following atoms has six valence electrons? - answer Sulfur

What is this structure for  $\text{H}_2$ ? - answer H-H

In a double bond, two electrons share a total of \_\_\_\_\_ electrons. - answer four

How many moles of HCl are present in 0.70L of a 0.33 M HCl solution - answer 0.23mol

What is the molarity of a solution that contains 125 g NaCl in 4.00 L solution? - answer 0.535M

How much KCl can you dissolve in 200g of 40 C water according to figure 1? - answer 80g

# Plato Chemistry Semester

**RS Peters**



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