6-8) Transforming Polynomial Functions

Example 1B: Translating a Polynomial Function

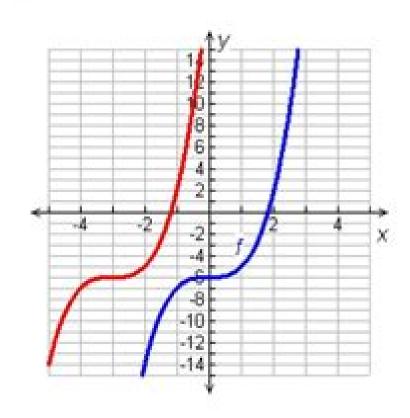
For $f(x) = x^3 - 6$, write the rule for each function and sketch its graph.

$$h(x) = f(x + 3)$$

 $h(x) = (x + 3)^3 - 6$

To graph h(x) = f(x + 3), translate the graph 3 units to the left.

This is a horizontal translation.



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Judd E. Hollander

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web 1 draw five protons in the nucleus of the atom label them with their charge 2 draw six neutrons in the nucleus of the atom 3 draw two electrons in the first energy level and label them with their charge 4 draw three electrons in the second energy level and label them with their charge 5 what element is represented by the diagram

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web may 18 2021 figure pageindex 1 the structure of the atom atoms have protons and neutrons in the center making the nucleus while the electrons orbit the nucleus the modern atomic theory states that atoms of one element are the same while atoms of different elements are different

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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as ts location on the periodic table no two different elements will have the give the symbol and number of protons in one atom of atomic number

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13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5 neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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