

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	58	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

World Of Chemistry Basic Review Sheet Answer

RS Peters



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Table of Contents World Of Chemistry Basic Review Sheet Answer

1. Understanding the eBook World Of Chemistry Basic Review Sheet Answer
 - The Rise of Digital Reading World Of Chemistry Basic Review Sheet Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying World Of Chemistry Basic Review Sheet Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an World Of Chemistry Basic Review Sheet Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from World Of Chemistry Basic Review Sheet Answer
 - Personalized Recommendations
 - World Of Chemistry Basic Review Sheet Answer User Reviews and Ratings
 - World Of Chemistry Basic Review Sheet Answer and Bestseller Lists
5. Accessing World Of Chemistry Basic Review Sheet Answer Free and Paid eBooks

- World Of Chemistry Basic Review Sheet Answer Public Domain eBooks
 - World Of Chemistry Basic Review Sheet Answer eBook Subscription Services
 - World Of Chemistry Basic Review Sheet Answer Budget-Friendly Options
6. Navigating World Of Chemistry Basic Review Sheet Answer eBook Formats
- ePub, PDF, MOBI, and More
 - World Of Chemistry Basic Review Sheet Answer Compatibility with Devices
 - World Of Chemistry Basic Review Sheet Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of World Of Chemistry Basic Review Sheet Answer
 - Highlighting and Note-Taking World Of Chemistry Basic Review Sheet Answer
 - Interactive Elements World Of Chemistry Basic Review Sheet Answer
8. Staying Engaged with World Of Chemistry Basic Review Sheet Answer
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers World Of Chemistry Basic Review Sheet Answer
9. Balancing eBooks and Physical Books World Of Chemistry Basic Review Sheet Answer
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection World Of Chemistry Basic Review Sheet Answer
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine World Of Chemistry Basic Review Sheet Answer
- Setting Reading Goals World Of Chemistry Basic Review Sheet Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of World Of Chemistry Basic Review Sheet Answer
- Fact-Checking eBook Content of World Of Chemistry Basic Review Sheet Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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