# SAMPLE PAPER - 2

Time: 3½ Hrs. MM: 100

#### GENERAL INSTRUCTIONS

- Question paper comprises of three sections, A, B and C you are to attempt all the three sections.
- II. All questions are compulsory.
- III. All questions of section A, section B and section C are to be attempted separately.
- IV. Question numbers 1 to 3 in section A are one mark each, to be answered in one word or one sentence.
- V. Question numbers 4 to 7 are two marks each, to be answered in about 30 words.
- VI. Question numbers 8 to 19 are three marks each, to be answered in about 50 words.
- VII. Question numbers 20 to 24 are five marks each, to be answered in about 70 words.
- VIII. Question numbers 25 to 42 in section B are MCQs based on practical skills. Each question is a one mark question.
- IX. Question numbers 43 and 44 in section C are OTBA based questions. Each question is of five marks. The open text material for these questions has been provided on pages (A-19-A-24)

# SECTION - A

- What conclusion would you derive from the observation, that in Rutherford's scattering experiment, most of the alpha-partieles passed through gold foil without any deflection?
- Name the organism in which asexual reproduction takes place by budding.
- Define power.
- 4. What are the advantages of classifying organisms?
- Acute diseases are severe but are less harmful than chronic diseases. Why?
- 6. Why a sound cannot be heard on the moon?
- 7. How is thrust different from pressure?
- The relative atomic mass of boron is 10.8.
  Calculate the percentage of its isotopes <sup>10</sup><sub>5</sub>B and <sup>11</sup><sub>5</sub>B occurring in nature.
- How many moles of Al atoms are needed to combine with 1.58 mole of O atoms to make aluminium oxide, Al<sub>2</sub>O<sub>3</sub>?
- 10. Copper oxide was prepared by the following methods:
  - (a) In one case, 1.75 g of the metal were dissolved in nitric acid and igniting the

- residual copper nitrate yielded 2.19 g of copper oxide.
- (b) In the second case, 1.14 g of metal dissolved in nitric acid were precipitated as copper hydroxide by adding caustic alkali solution. The precipitated copper hydroxide after washing, drying and heating yielded 1.43g of copper oxide.
- (c) In the third case, 1.45 g of copper when strongly heated in a current of air yielded 1.83 g of copper oxide. Show that the given data illustrate the law of constant composition.
- 11. Why are the protozoan placed in kingdom-Protista and not in kingdom-Animalia?
- Classify the following organisms based on the absence/presence of true coelom (i.e., acoelomate, pseudocoelomate and coelomate) Spongilla, Sea anaemone, Planaria, Liver fluke Wichereria, Ascaris, Nereis, earthworm, Scorpion, Birds, Fishes, horse.
- 13. What is antibiotic Penicillin? Give its function.
- 14. List any two differences between transverse and longitudinal waves. What is the nature of sound waves in air?

# **Sample Paper For Class 9 Sa2 Science**

**Petre Tautu** 

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