

General Chemistry placement test Study Guide 2024 Solved 100%

Atomic Number - The number of protons in the nucleus of an atom. The atomic number of an element never changes, meaning that the number of protons in the nucleus of every atom in an element is always the same.

Hydrogen = 1, Carbon = 6, Oxygen = 8 (located on top of element of periodic table)

Mass Number - # of neutrons + # of protons = mass number (located on bottom of periodic table)

Isotopes - Let's say an atom is missing a neutron or has an extra neutron. That type of atom is called an isotope. An atom is still the same element if it is missing an electron.

Example: (carbon-14)

Distribution B/W subatomic particles in an Atom - A typical atom consists of three subatomic particles: Protons, Neutrons, and Electrons.

- Nucleons include protons and neutrons. All the positive charge of an atom is contained in the nucleus, and originates from the protons. Neutrons are neutrally-charged. Electrons, which are negatively-charged, are located outside of the nucleus.

Roles of subatomic particles: Protons, Neutrons, & Electrons - **Particle Mass:**

Protons & Neutrons have about same mass. Electrons have much less mass than protons and neutrons.

Particle Charge:

Protons have a positive charge (+)

- Neutrons have no charge (0)

- Net charge of the nucleus is positive and equal to the number of protons.

Texas Tech Chemistry Placement Exam Study Guide

Thomas Griffiths



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In a global inundated with screens and the cacophony of quick transmission, the profound power and psychological resonance of verbal artistry often disappear in to obscurity, eclipsed by the regular onslaught of noise and distractions. Yet, situated within the musical pages of **Texas Tech Chemistry Placement Exam Study Guide**, a interesting work of literary splendor that pulses with natural feelings, lies an wonderful trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, that interesting opus guides visitors on a mental odyssey, gently revealing the latent possible and profound affect stuck within the elaborate internet of language. Within the heart-wrenching expanse of the evocative analysis, we shall embark upon an introspective exploration of the book is main subjects, dissect their charming writing design, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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 $5co i2 5co2 i 2 o 5 5 c o i 2 5 c o 2$

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