Equation Writing and Predicting Products

he treetesctions:

If you examine a car that has been in a junkyard for a while, you will notice that it has rusted. Rusting in a slow chemical reaction of the iron to the ear's body with oxygen gas. If a piece of sodium is put into water, a much more rapid chemical reaction occurs. Sodium reacts with water to produce sodium bydroxide and bydrogen gas. During this reaction, enough heat is liberated to ignite the hydrogen gas, causing it to explode.

Chemists observe what is happening in a chemical reaction and try to describe it in language that is simple and clear. A chemical equation uses formulas and symbols to describe the substances involved in a reaction, the physical state of the substance, the use of a catalyst, and relative proportions. The general form of an equation is:

Reactants ---- Preducts

In this investigation you will perform a series of reactions and make careful observations of the changes that occur. Using simple tests and your knowledge of chemistry, you will describe the identity of the products. With this information, you will write chemical equations to describe the reactions.

Pre-Eab Discussion:

Read firesigh the entire lab investigation and the relevant pages of your textbook. Then answer the questions that killow.

- 1. What constitutes a positive test for each of the following gases?
 2. expgen
 3. byptrogen
 4. water vapor
 6. structure (NFL)

 2. What is the proper way to smell a substance in the lab? Why should care be taken when smelling a gas such as ammonia?

 3. What is the role of a satisfyst in a reaction? How can yes tell when a substance serves as a catalyst? (This answer will be important later, spend some time on it now.)

 4. One way to identify limestone (CaCO₄) is to drop a small amount of hydrochionic acid (HCT) on it. A positive test results in a first of carbon districts being
 - produced. Write out in words the information represented by the following balanced element equation.

 CaCO₂(s) + 2 HCI (asp. —— CaCU₂(sq) + H₂O(l) + CO₂(g).

Predicting Products And Writing Equations Lab 25

JE Gale

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Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02 Prentice Hall Chemistry ,2000 Foundations of College Chemistry, Alternate Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They Il learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis **Instructor's Guide for Introductory Chemistry in the Laboratory** James F. Hall, 1996 Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry. An emphasis on real world topics lets readers clearly see how the Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning chemistry will apply to their career the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

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Geometry Cord, 2000 Holt Chemistry Ralph Thomas Myers, 2004 **Scientific and Technical Aerospace Reports** ,1983 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database Research Abstracts, 1988 **ERDA Energy Research Abstracts** ,1983 **Bridges to Algebra and Geometry** ,2004 Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Microbiology Abstracts ,1990 Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations Cities and Their Vital Systems Advisory Committee on Technology and Society, 1989 Cities and Their Vital Systems asks basic questions about the longevity utility and nature of urban infrastructures analyzes how they grow interact and change and asks how when and at what cost they should be replaced Among the topics discussed are problems arising from increasing air travel and airport congestion the adequacy of water supplies and waste treatment the impact of new technologies on construction urban real estate values and the field of telematics the combination of computers and telecommunications that makes money machines and national newspapers possible

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