

## 14

## THE BEHAVIOR OF GASES

## Chapter Test II

## A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct answer next to the item.

## Column I

1. The constant relating unit temperature changes to changes in volume for a fixed amount of gas at constant pressure (the ratio of the change in volume to the change in temperature).
2. The volume of a fixed amount of a gas at constant pressure in the standard state (0°C and 1 atm).
3. The ratio of the mass of a gas to its volume, independent of the actual amount of gas used.
4. An experimental gas constant determined from the ideal gas law at low pressures.
5. A constant of nature that has values of about 8.314 J/K·mol.
6. The a parameter of gas constant equations for determining the gas molecular weight of a gas.
7. An experimental gas constant derived from the ideal gas law under conditions considered to be ideal.
8.  $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. The amount of heat enough to raise 1 g of water 1°C.
10. A unit of energy (joules).

## Column II

- a. standard gas law
- b. ideal gas constant R
- c. volume
- d. compressibility
- e. Boyle's law
- f. molar volume
- g. standard molar volume
- h. density
- i. compressibility
- j. heat capacity

## B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. According to Graham's law, the rate of effusion of a gas is proportional to the square root of its molecular weight. Which of the following gases effuses most rapidly?  
a. the heaviest gas  
b. the lightest gas  
c. the gas with the highest molecular weight  
d. the gas with the lowest molecular weight

# Prentice Hall Chemistry Ch 14 Study Guide

**John McMurry, Susan McMurry**



## **Prentice Hall Chemistry Ch 14 Study Guide:**

**Prentice Hall Chemistry**, 2000     *Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry* John McMurry, Susan McMurry, 1986     *Study Guide and Solutions Manual for Organic Chemistry* Susan McMurry, 1992 John McMurry's best selling text presents organic chemistry in a new edition that is up to date beautifully written visually striking and pedagogically sound Described by many of its users as an eminently teachable text McMurry sets the standard in the field The writing style has received almost universal acclaim from its users McMurry introduces new concepts only as needed and immediately illustrates them with concrete examples And wherever possible he ties material together with brief reviews overviews and reaction summaries The result is a text that helps students mentally organize the material a text that helps them understand concepts not just memorize facts and a text that helps them make sense of the voluminous amount of material they encounter in the study of organic chemistry McMurry uses a simple but important polar reaction the addition of HBr to an alkene as the lead off reaction to illustrate the general principles of organic reactions Users of former editions found this an excellent choice because of its relative simplicity no prior knowledge of chirality or kinetics is required and its importance as a polar reaction on a common functional group that offers students the key to understanding hundreds of thousands of ionic reactions By selecting this particular model McMurry is able to offer an unusually early presentation of organic reactions     Introduction to Materials Chemistry Harry R. Allcock, 2019-10-08 This textbook introduces the reader to the elementary chemistry on which materials science depends by discussing the different classes of materials and their applications It shows the reader how different types of materials are produced why they possess specific properties and how they are used in technology Each chapter contains study questions to enable discussions and consolidation of the acquired knowledge The new edition of this textbook is completely revised and updated to reflect the significant expansion of the field of materials chemistry over the last years covering now also topics such as graphene nanotubes light emitting diodes extreme photolithography biomedical materials and metal organic frameworks From the reviews of the first edition This book is not only informative and comprehensive for a novice reader but also a valuable resource for a scientist and or an industrialist for new and novel challenges Materials and Manufacturing Process June 2009 Allcock provides a clear path by first describing basic chemical principles then distinguishing between the various major materials groups and finally enriching the student by offering a variety of special examples CHOICE April 2009 Proceeding logically from the basics to materials in advanced technology it covers the fundamentals of materials chemistry including principles of materials synthesis and materials characterization methods Internationale Fachzeitschrift Metall January 2009 *Teaching Chemistry - A Studybook* Ingo Eilks, Avi Hofstein, 2013-04-20 This book focuses on developing and updating prospective and practicing chemistry teachers pedagogical content knowledge The 11 chapters of the book discuss the most essential theories from general and science education and in the second part of each of the chapters apply the theory to

examples from the chemistry classroom Key sentences tasks for self assessment and suggestions for further reading are also included The book is focused on many different issues a teacher of chemistry is concerned with The chapters provide contemporary discussions of the chemistry curriculum objectives and assessment motivation learning difficulties linguistic issues practical work student active pedagogies ICT informal learning continuous professional development and teaching chemistry in developing environments This book with contributions from many of the world s top experts in chemistry education is a major publication offering something that has not previously been available Within this single volume chemistry teachers teacher educators and prospective teachers will find information and advice relating to key issues in teaching such as the curriculum assessment and so forth but contextualised in terms of the specifics of teaching and learning of chemistry and drawing upon the extensive research in the field Moreover the book is written in a scholarly style with extensive citations to the literature thus providing an excellent starting point for teachers and research students undertaking scholarly studies in chemistry education whilst at the same time offering insight and practical advice to support the planning of effective chemistry teaching This book should be considered essential reading for those preparing for chemistry teaching and will be an important addition to the libraries of all concerned with chemical education Dr Keith S Taber University of Cambridge Editor Chemistry Education Research and Practice The highly regarded collection of authors in this book fills a critical void by providing an essential resource for teachers of chemistry to enhance pedagogical content knowledge for teaching modern chemistry Through clever orchestration of examples and theory and with carefully framed guiding questions the book equips teachers to act on the relevance of essential chemistry knowledge to navigate such challenges as context motivation to learn thinking activity language assessment and maintaining professional expertise If you are a secondary or post secondary teacher of chemistry this book will quickly become a favorite well thumbed resource Professor Hannah Sevian University of Massachusetts Boston

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