

## Chapter 7 Chemical Reactions

**Section 7.1 Describing Reactions**

(pages 192–198)

*This section discusses the use of chemical equations and how to balance them. It also demonstrates the use of calculations in chemistry.*

**Reading Strategy (page 192)**

**Monitoring Your Understanding** Preview the Key Concepts, topic headings, vocabulary, and figures in this section. List two things you expect to learn. After reading, state what you learned about each item you listed. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Expect to Learn	What I Learned
Answers may vary. Possible answers: How to balance chemical equations	Answers may vary. Possible answers: An unbalanced equation can be balanced by changing the coefficients.
How to convert from mass to moles	The mass of a substance can be converted to moles by using the molar mass as a conversion factor.

**Chemical Equations (pages 192–193)**

- Is the following sentence true or false? The new substances formed as a result of a chemical reaction are called products. true
- Circle the letter of each sentence that is a correct interpretation of the chemical equation  $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$ .
  - Carbon and oxygen react and form carbon monoxide.
  - ☒ Carbon and oxygen react and form carbon dioxide.
  - Carbon dioxide yields carbon and oxygen.
  - ☒ The reaction of carbon and oxygen yields carbon dioxide.
- Is the following sentence true or false? The law of conservation of mass states that mass is neither created nor destroyed in a chemical reaction. true
- Circle the letter of the correct answer. According to the equation  $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$ , how many carbon atoms react with 14 molecules of oxygen to form 14 molecules of carbon dioxide?
  - 1
  - 7
  - ☒ 14
  - 28
- In the reaction represented by the equation  $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$ , the mass of carbon dioxide produced equals the total mass of carbon and oxygen that reacted.

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Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02

Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30

With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum

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**Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition** John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

**Fundamentals of Nursing E-Book** Barbara L Yoost, Lynne R Crawford, 2019-01-23 Yoost and Crawford's Fundamentals of Nursing is back for a second edition encore The text that made its name by focusing on simple language and active learning continues its focus on helping you truly understand apply and retain important nursing information and concepts Using a warm and conversational style this new second edition guides you towards a basic understanding of the nursing profession and then logically progresses through the nursing process and into the safe and systematic methods of applying care Each chapter features realistic and complex case studies and critical thinking exercises woven throughout the content to help you continually apply what you've learned to actual patient care A conceptual care mapping approach created by Yoost and Crawford themselves further your ability to make clinical judgments and synthesize knowledge as you develop plans of care after analyzing and clustering related patient assessment data All of this paired with a wealth of student friendly learning features and clinically focused content offers up a fundamentally different and effective way for you to easily master the fundamentals of nursing

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**Biodeterioration of Concrete** Thomas Dyer,2017-07-12 Awareness of the importance of ensuring durability of concrete has been a growing concern of engineers and there is now considerable understanding of the mechanisms which cause its deterioration and means of limiting such damage through the use of appropriate materials and approaches to design Many of the deterioration mechanisms which affect concrete are the result of interaction with the non living environment chlorides in seawater carbon dioxide in the atmosphere cyclic freezing and thawing However living organisms can also cause damage through both chemical and physical processes which under the right conditions can be severe This book looks at all forms of concrete biodeterioration together for the first time It examines from a fundamental starting point biodeterioration mechanisms as well as the conditions which allow living organisms bacteria fungi plants and a range of marine organisms to colonise concrete A detailed evaluation of chemical compounds produced by living organisms with respect to their interaction with the mineral constituents of concrete and the implications it has for the integrity of structures is also included Approaches to avoiding biodeterioration of concrete are also covered including selection of materials mix proportioning design and use of protective systems

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