

11-2 Practice Problems: Mass-spectrometry, empirical and molecular formulae

1. Calculate percentage composition of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.
2. Calculate the empirical formula of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.
3. Calculate the percentage composition of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.
4. Calculate the empirical formula of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.
5. Calculate the percentage composition of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.
6. Calculate the percentage composition of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.

Empirical and Molecular Formulae

1. Calculate the empirical formula of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.
2. Calculate the empirical formula of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.
3. Calculate the empirical formula of a hydrocarbon mixture. A hydrocarbon with C and H only, with a molar mass of 58.12 g/mol.

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TD Snyder



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