

Chemistry 105

Chapter 3 Review Problems

1. How many molecules are present in 180 grams of water? (The atomic weights of H and O are 1 g/mol and 16 g/mol, respectively.)
a) 6.0×10^{24}
b) 10
c) 5.6×10^{-23}
d) 1.8×10^7
e) 180×10^{23}
2. The atomic mass of sulfur is given as 32.07 g/mol. What is the mass of an average atom of sulfur?
a) 5.33×10^{-23} g
b) 32.07 g
c) 32.07×10^{-23} kg
d) 193×10^{-22} g
e) 6.02×10^{23} g
3. What is the mass of C present in 60 g of C_2H_6 ?
a) 12 g
b) 24 g
c) 48 g
d) 60 g
e) 6 g
4. What is the molar mass of C_2H_6O ?
a) 24 g/mol
b) 32 g/mol
c) 7 g/mol
d) 35 g/mol
e) 46 g/mol
5. What is the correct percentage by mass of nitrogen in HCN?
a) 33 %
b) 14%
c) 27%
d) 52%
e) 98%
6. What is the chief danger that is anticipated if CO_2 levels get too high in the atmosphere?
a) people will develop skin cancer from exposure to UV light
b) the atmosphere will develop an unpleasant odor
c) the temperature of the planet may climb by several degrees
d) plant life will be exterminated
e) The CO_2 will remove ozone from the stratosphere

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weapons of mass destruction The text is also the only book on the market with a detailed description of DNA and CODIS techniques used by professionals

Mass and Energy Balances in Materials Engineering Mark A. Schlesinger, 1996 This text takes a practical approach to its presentation of stoichiometry and energy balance principles for materials engineering students by emphasizing their use in actual engineering practice

British Chemical Engineering & Process Technology, 1966

Molecular Modeling of Corrosion Processes Christopher D. Taylor, Philippe Marcus, 2015-05-26 Presents opportunities for making significant improvements in preventing harmful effects that can be caused by corrosion Describes concepts of molecular modeling in the context of materials corrosion Includes recent examples of applications of molecular modeling to corrosion phenomena throughout the text Details how molecular modeling can give insights into the multitude of interconnected and complex processes that comprise the corrosion of metals Covered applications include diffusion and electron transfer at metal electrolyte interfaces Monte Carlo simulations of corrosion corrosion inhibition interrogating surface chemistry and properties of passive films Presents current challenges and likely developments in this field for the future

American Scientist, 1942

A Complete Preparation for the MCAT Aftab S. Hassan, Dorothy Haberkamp Air, 1997

Robotica, 1987

Principles of Inorganic Materials Design John N. Lalena, David A. Cleary, Olivier B.M. Hardouin Duparc, 2020-05-27 Learn the fundamentals of materials design with this all inclusive approach to the basics in the field Study of materials science is an important aspect of curricula at universities worldwide This text is designed to serve students at a fundamental level positioning materials design as an essential aspect of the study of electronics medicine and energy storage Now in its 3rd edition Principles of Inorganic Materials Design is an introduction to relevant topics including inorganic materials structure property relations and material behaviors The new edition now includes chapters on computational materials science intermetallic compounds and covalent compounds The text is meant to aid students in their studies by providing additional tools to study the key concepts and understand recent developments in materials research In addition to the many topics covered the textbook includes Accessible learning tools to help students better understand key concepts Updated content including case studies and new information on computational materials science Practical end of chapter exercises to assist students with the learning of the material Short biographies introducing pioneers in the field of inorganic materials science For undergraduates just learning the material or professionals looking to brush up on their knowledge of current materials design information this text covers a wide range of concepts research and topics to help round out their education The foreword to the first edition was written by the 2019 Chemistry Nobel laureate Prof John B Goodenough

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