

Sample Question Paper

4

Chemistry

A sample question paper for **CBSE** Class XI

Time : 3 hrs

Max. Marks : 70

General *Instructions*

1. All questions are compulsory.
2. The question paper consists of 26 questions divided into five sections A, B, C, D and E. Section A comprises of 5 questions of 1 mark each, Section B comprises of 5 questions of 2 marks each, Section C comprises of 12 questions of 2 marks each, Section D comprises of 1 questions of 4 marks and Section E comprises of 3 questions of 5 marks each.
3. All questions in Section A are to be answered in one word, one sentence or as per the exact requirement of the question.
4. There is no overall choice. However, internal choice has been provided in some questions. You have to attempt only one of the alternative in all such questions.
5. Use of calculator is not permitted. You may ask for logarithmic table, if required.

Section A

1. Is the law of constant composition true for all types of compounds? Mention why or why not?
2. What was the necessity for the classification of the elements?
3. Water decomposes by absorbing 286.2 kJ of electrical energy per mole. When H_2 and O_2 combine to form one mole of H_2O , 286.2 kJ of heat is produced. Which law is proved? What statement of the law follows from it?
4. Which d-orbital does not have four lobes and what is its shape called?
5. Which species is the smallest aromatic substance?

Section B

6. What is hybrid state of central atom in the following?

NO_3^- , BF_4^- , PF_5 , IF_5 and CO_2

Sample Questions Paper

Siddappa N.Byrareddy



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Table of Contents Sample Questions Paper

1. Understanding the eBook Sample Questions Paper
 - The Rise of Digital Reading Sample Questions Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Questions Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Questions Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Questions Paper
 - Personalized Recommendations
 - Sample Questions Paper User Reviews and Ratings
 - Sample Questions Paper and Bestseller Lists
5. Accessing Sample Questions Paper Free and Paid eBooks
 - Sample Questions Paper Public Domain eBooks
 - Sample Questions Paper eBook Subscription Services
 - Sample Questions Paper Budget-Friendly Options
6. Navigating Sample Questions Paper eBook Formats

- ePub, PDF, MOBI, and More
- Sample Questions Paper Compatibility with Devices
- Sample Questions Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Questions Paper
 - Highlighting and Note-Taking Sample Questions Paper
 - Interactive Elements Sample Questions Paper
- 8. Staying Engaged with Sample Questions Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Questions Paper
- 9. Balancing eBooks and Physical Books Sample Questions Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Questions Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sample Questions Paper
 - Setting Reading Goals Sample Questions Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sample Questions Paper
 - Fact-Checking eBook Content of Sample Questions Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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