

## 2-4: Precipitation Reactions

1. Start *Virtual ChemLab*, select *Reactions and Stoichiometry*, and then select *Precipitation Reactions* from the list of assignments. The lab will open in the Inorganic laboratory.
2. React each of the cations (across the top) with each of the anions (down the left) according to the data table using the following procedures:

**Data Table**

|                                                 | $\text{AgNO}_3$ ( $\text{Ag}^+$ ) | $\text{Pb}(\text{NO}_3)_2$ ( $\text{Pb}^{2+}$ ) | $\text{Ca}(\text{NO}_3)_2$ ( $\text{Ca}^{2+}$ ) |
|-------------------------------------------------|-----------------------------------|-------------------------------------------------|-------------------------------------------------|
| $\text{Na}_2\text{CO}_3$ ( $\text{CO}_3^{2-}$ ) | <i>a</i>                          | <i>f</i>                                        | <i>k</i>                                        |
| $\text{Na}_2\text{S}$ ( $\text{S}^{2-}$ )       | <i>b</i>                          | <i>g</i>                                        | <i>l</i>                                        |
| $\text{NaOH}$ ( $\text{OH}^-$ )                 | <i>c</i>                          | <i>h</i>                                        | <i>m</i>                                        |
| $\text{Na}_2\text{SO}_4$ ( $\text{SO}_4^{2-}$ ) | <i>d</i>                          | <i>i</i>                                        | <i>n</i>                                        |
| $\text{NaCl}$ ( $\text{Cl}^-$ )                 | <i>e</i>                          | <i>j</i>                                        | <i>o</i>                                        |

- a. Drag a test tube from the box and place it on the metal test tube stand. You can then click on the bottle of  $\text{Ag}^+$  ion solution on the shelf to add it to the test tube.
  - b. Click on the **Divide** button on the edge of the lab bench four times to make four additional test tubes containing  $\text{Ag}^+$ . With one test tube in the metal stand and four others in the blue rack, click on the  $\text{Na}_2\text{CO}_3$  bottle on the reagent shelf and observe what happens in the window at the top left. Record your observation in the table above. If the solution remains clear, record NR for no reaction. Drag this test tube to the red disposal bucket on the right.
  - c. Drag a second tube from the blue rack to the metal stand. Add  $\text{Na}_2\text{S}$ , record your observations and discard the tube. Continue with the third, fourth and fifth tube, but add  $\text{NaOH}$ ,  $\text{Na}_2\text{SO}_4$ , and  $\text{NaCl}$  respectively. Record your observations and discard the tubes. When you are finished, click on the red disposal bucket to clear the lab.
  - d. Repeat steps a-c for five test tubes of  $\text{Pb}^{2+}$  and  $\text{Ca}^{2+}$ . Record your observations in the data table. If no precipitate forms write NR for no reaction.
3. What happens in grid space *d*? What other reactions give similar results? Is it necessary to write an equation when no reaction occurs? Explain. \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
  4. Write balanced equations for all precipitation reactions you observed.
    - a.
    - b.
    - c.

# Precipitation Reactions Virtual Chemlab Answers

**Brian F. Woodfield, Matthew C.  
Asplund, Steven Haderlie**



## **Precipitation Reactions Virtual Chemlab Answers:**

Virtual Chemlab Brian F. Woodfield, Matthew C. Asplund, Steven Haderlie, 2006 Contains a full virtual lab environment as well as the pre arranged labs that are referenced in the workbook and at the end of the chapter in the textbook Virtual ChemLab can be run directly from the CD or installed on the student's computer **Instructor's Manual** Brian F. Woodfield, Matthew C. Asplund, 2006 NEW Click here to visit the Virtual ChemLab Frequently Asked Questions FAQ document This Instructor's Lab Manual Workbook is similar to the Student Lab Manual Workbook and additionally contains an overview of the full capabilities of the Site License version of Virtual ChemLab installation instructions and the answers for the laboratory assignments provided in the student laboratory workbook This product is available within Virtual ChemLab General Chemistry Instructor Lab Manual Workbook and Student CD Combo Package v2 5 0 13 228010 8 Valuepack and or should be ordered in conjunction with Virtual ChemLab General Chemistry Instructor Site License CD v2 5 0 13 185749 5

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*Reactions in water solutions* Khalil H. Mancy, Frederick G. Pohland, 1973 **Kinetics of Precipitation Reactions** Pedro Eduardo José Rivera Díaz del Castillo, 2002

## The Enigmatic Realm of **Precipitation Reactions Virtual Chemlab Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Precipitation Reactions Virtual Chemlab Answers** a literary masterpiece penned with a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of people who partake in its reading experience.

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