After these lessons, I hope you have a better understanding of nuclear physics.

In your notes on nuclear physics, please fill in the blanks under Section IV Radioactive Decay as follows:

1. Alpha Decay

$$^{228}_{90}\text{Th} \rightarrow {}^{4}_{2}\text{He} + {}^{224}_{88}\text{Ra}$$

2. Beta Decay

The unstable nucleus becomes stable by converting a neutron into a proton and an electron. The electron is then ejected from the nucleus at high speed.

$$^{234}_{90}\text{Th} \rightarrow ^{0}_{-1}\text{e} + ^{234}_{91}\text{Pa}$$

3. Gamma Emission

After emitting an α or β particle, the remaining nuclei may have surplus energy. As a result, γ radiation may be emitted.

$$^{234}_{90}\text{Th*} \rightarrow ^{234}_{90}\text{Th} + \gamma$$

Predicting Nuclear Decay Answers

Puja Mehta

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