

THE NATURE OF GASES

(p1)

review

18 pts
- 2 pt is 1st
- 1 pt is 2nd 3rd

- Describe the assumptions of the kinetic theory as it applies to gases.
- Express gas pressure in terms of kinetic theory.
- Define the relationship between Kelvin temperature and average kinetic energy.

Vocabulary

- kinetic energy
- kinetic theory
- gas pressure
- vacuum
- atmospheric pressure
- barometer

Part A Completion

Use this completion exercise to check your understanding of the concepts that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

The kinetic theory describes the 1 of particles in matter and the forces of attraction between them. The theory assumes that the volume occupied by a gas is mostly 2, that the particles of gas are relatively 3, move 4 of each other, and are in constant 5 motion. The 6 between particles are perfectly elastic so that the total 7 remains constant. Gas pressure results from the simultaneous collisions of billions of particles with an object. Barometers are used to measure 8 pressure. Standard conditions are defined as a temperature of 9 and a pressure of 10.

- motion
- empty space
- far apart
- independently
- random / rapid
- collisions
- kinetic energy
- atmospheric
- 0°C / 273 K
- 1 atm / 101.3 kPa

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- ST 11. Atmospheric pressure is 760 mm Hg.
AT 12. The SI unit of pressure is the pascal.

↳ a multiple of pascal

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Prentice Hall Chemistry Chapter 13 Assessment Answers

David R. Hittle



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Multiphase Particulate Systems in Turbulent Flows Wioletta Podgórska, 2019-09-17 Multiphase Particulate Systems in Turbulent Flows Fluid Liquid and Solid Liquid Dispersions provides methods necessary to analyze complex particulate systems and related phenomena including physical chemical and mathematical description of fundamental processes influencing crystal size and shape suspension rheology interfacial area of drops and bubbles in extractors and bubble columns Examples of mathematical model formulation for different processes taking place in such systems is shown Discussing connections between turbulent mixing mechanisms and precipitation it discusses influence of fine scale structure of turbulence including its intermittent character on breakage of drops bubbles cells plant cell aggregates An important aspect of the mathematical modeling presented in the book is multi fractal taking into account the influence of internal intermittency on different phenomena Key Features Provides detailed descriptions of dispersion processes in turbulent flow interactions between dispersed entities and continuous phase in a single volume Includes simulation models and validation experiments for liquid liquid gas liquid and solid liquid dispersions in turbulent flows Helps reader learn formulation of mathematical models of breakage or aggregation processes using multifractal theory Explains how to solve different forms of population balance equations Presents a combination of theoretical and engineering approaches to particulate systems along with discussion of related diversity with exercises and case studies Fluid Chemistry, Drilling and Completion Qiwei Wang, 2021-11-04 Fluid Chemistry Drilling and Completion the latest release in the Oil and Gas Chemistry Management series that covers all sectors of oil and gas chemicals from drilling to production processing storage and transportation delivers critical chemical oilfield basics while also covering the latest research developments and practical solutions Organized by type of chemical the book allows engineers to fully understand how to effectively control chemistry issues make sound decisions and mitigate challenges Sections cover downhole sampling crude oil characterization such as fingerprinting

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Principles of Environmental Chemistry Roy M. Harrison,Royal Society of Chemistry (Great Britain),2007 Environmental chemistry is becoming increasingly important and is crucial in the understanding of a range of issues ranging from climate change to local pollution problems Principles of Environmental Chemistry draws upon sections of the authors previous text Understanding our Environment and reflects the growing trend of a more sophisticated approach to teaching environmental science at university This new revised text book focuses on the chemistry involved in environmental problems Written by leading experts in the field the book provides an in depth introduction to the chemical processes influencing the atmosphere freshwaters salt waters and soils Subsequent sections discuss the behaviour of organic chemicals in the environment and environmental transfer between compartments such as air soil and water Also included is a section on biogeochemical cycling which is crucial in the understanding of the behaviour of chemicals in the environment Complete with worked examples the book is aimed at advanced undergraduate and graduate chemistry students studying environmental chemistry

Resources in education ,1984-09 *Chemical Property Estimation* Edward Baum,2018-05-11 Our world is widely contaminated with damaging chemicals and companies create thousands of new potentially dangerous chemicals each year Due to the difficulty and expense of obtaining accurate measurements and the unreliability of reported values we know surprisingly little about the properties of these contaminants Determining the properties of chemicals is critical to judging their impact on environmental quality and in making decisions about emission rates clean up and other important public health issues Chemical Property Estimation describes modern methods of estimating chemical properties methods which cost much less than traditional laboratory techniques and are sufficiently accurate for most environmental applications Estimation methods are used to screen chemicals for testing design

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Criminalistics: Forensic Science, Crime, and Terrorism James E. Girard, 2013-11-13 Designed for students that are not biology chemistry or physics majors this fully revised and updated Third Edition of the best selling Criminalistics Forensic Science Crime and Terrorism provides a comprehensive introduction to forensic science the scientific principles that are the underpinnings of crime analysis and the practical application of these principles Essential topics such as fingerprint identification DNA ballistics detection of forgeries forensic toxicology computer forensics and the identification and analysis of illicit drugs are thoroughly explained in a reader friendly manner Unlike comparable texts the Third Edition includes coverage of important terrorism and homeland security issues including explosives cybercrime cyberterrorism and weapons of mass destruction The text is also the only book on the market with a detailed description of DNA and CODIS techniques used by professionals

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Organic Chemistry Herman Harry Szmant, 1957 **Chemistry** James C. Hill, 2003 This book assists students through the text material with chapter overviews learning objectives review of key terms cumulative chapter review quizzes and self tests Included are answers to all Student Guide exercises Chapter summaries are correlated to those in the Instructor's Resource Manual

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