

Anions, Cations, and Ionic Reactions

Hands-On Labs, Inc.

Version 42-0304-00-01

Lab Report Assistant

This document is not meant to be a substitute for a formal laboratory report. The Lab Report Assistant is simply a summary of the experiment's questions, diagrams if needed, and data tables that should be addressed in a formal lab report. The intent is to facilitate students' writing of lab reports by providing this information in an editable file which can be sent to an instructor.

Exercise 1: Qualitative Anion Tests

Data Table 1. Anion Confirmation Tests.

| Chemical | Anion | Observations: Addition of AgNO_3 | Observations: Addition of HCl |
|--------------------------|-----------|---|--|
| NaBr | Bromide | Turned milky white | Turned yellowish More of a gel |
| Na_2CO_3 | Carbonate | Thick white glob A little grainy | White sediment in a clear liquid Bubbles up |
| NaCl | Chloride | Milky white | White sediment visible |
| NaI | Iodide | Yellow glob immediately when dropped in | Yellow, grainy/sediment, thick looking |
| Na_2SO_4 | Sulfate | Thickened | Very cloudy/milky |
| Na_2S | Sulfide | Brown Globbs immediately when dropped in Still watery | Brown globbs thinned out into brown pieces of sediment Still watery (brownish) looks like dirty water |
| Unknown #103 | Carbonate | Thick white glob Turned gel | Bubbly and cloudy Very thick gel |
| Unknown #104 | Sulfide | Brown globbs immediately when dropped in Still watery | Brown globbs thinned out into brown pieces of sediment Still watery (brownish) looks like dirty water |

Sample Lab Report For Reactions Of Anions

E Durkheim



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