Stoichiometry Calculation Practice Worksheet

 Calculate the number of moles of NaOH that are needed to react with 500.0 g of H₂SO₄ according to the following equation:

$$H_2SO_4 + 2 NaOH \rightarrow Na_2SO_4 + 2 H_2O$$

 Calculate the mass of NH₃ that can be produced from the reaction of 125 g of NCl₃ according to the following equation:

$$NCl_3 + 3 H_2O \rightarrow NH_3 + 3 HOC1$$

 What mass of CO₂ can be produced from the reaction of 25.0 g of C₃H₈ with 75.0 g of O₂ according to the following equation:

$$C_3H_8$$
 + 5 O_2 \rightarrow 3 CO_2 + 4 H_2O

Stoichiometry 11 3 Practice Problems

Kathleen Armour

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