

Stoichiometry, Limiting Reactants, Percent Yield, and Excess Reactant Remaining Worksheet

1. **Aluminum** reacts with **oxygen** to produce **aluminum oxide** which can be used as an adsorbent, desiccant or catalyst for organic reactions.

1 mole of aluminum (Al) is 26.982 g
1 mole of molecular oxygen (O₂) is 31.998 g
1 mole of aluminum oxide is 101.961 g

- a. What is the formula of aluminum oxide? (First hint: use the criss-cross rule with aluminum ion and oxide ion) (Second hint: check your answer with the molar mass of aluminum oxide)



- b. Write the balanced chemical equation for the reaction of aluminum (Al) with molecular oxygen (O₂) to make aluminum oxide.



- c. How many grams of aluminum oxide could form if 82.49 g Al reacts completely?

$$82.49 \text{ g Al} \times \frac{1 \text{ mol Al}}{26.982 \text{ g Al}} \times \frac{2 \text{ mol Al}_2\text{O}_3}{4 \text{ mol Al}} \times \frac{101.961 \text{ g Al}_2\text{O}_3}{1 \text{ mol Al}_2\text{O}_3} = 155.9 \text{ g Al}_2\text{O}_3$$

- d. How many grams of aluminum oxide could form if 117.65 g O₂ reacts completely?

$$117.65 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{31.998 \text{ g O}_2} \times \frac{2 \text{ mol Al}_2\text{O}_3}{3 \text{ mol O}_2} \times \frac{101.961 \text{ g Al}_2\text{O}_3}{1 \text{ mol Al}_2\text{O}_3} = 249.9 \text{ g Al}_2\text{O}_3$$

Stoichiometry Limiting Reactant Percent Yield Quiz

Guillaume Favre



Stoichiometry Limiting Reactant Percent Yield Quiz:

Decoding **Stoichiometry Limiting Reactant Percent Yield Quiz**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Stoichiometry Limiting Reactant Percent Yield Quiz**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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