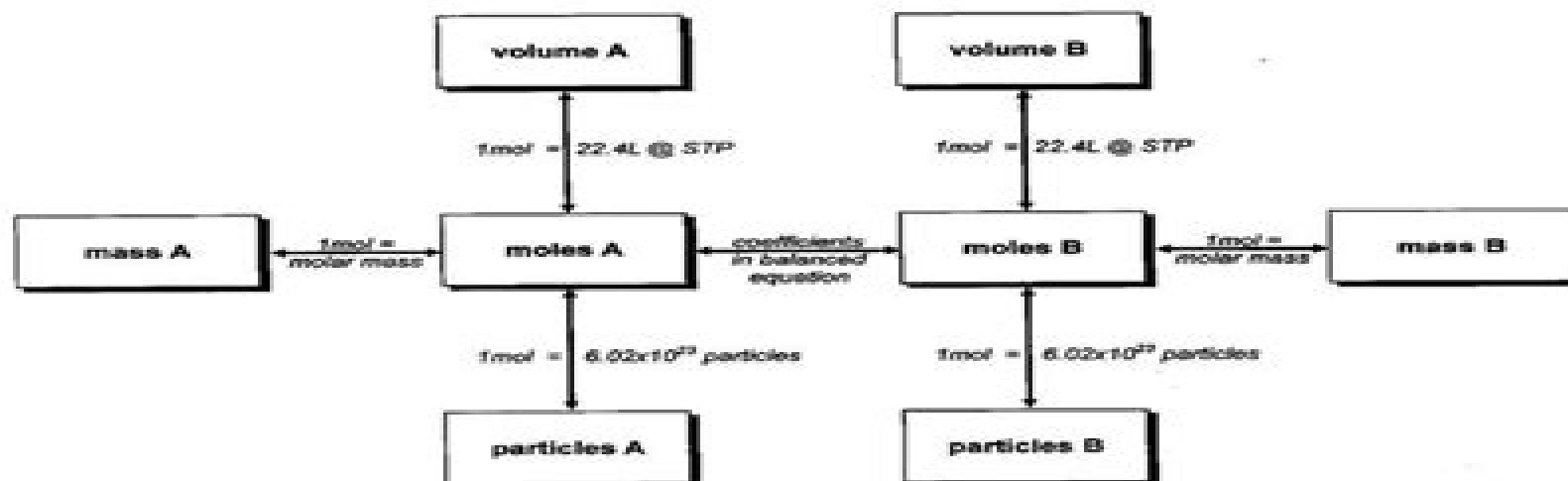


Stoichiometry 4: Mixed Problem Practice

Introduction

In this set of practice problems, you will convert from between a mixture of units (grams, moles, liters, and particles) within the substances of a chemical reaction. Use the flowchart below as a reminder of how to convert between the different units as needed.



Example

The carbon dioxide gas exhaled by the crew in a space shuttle is removed from the air by a reaction with canisters of lithium hydroxide, producing a solution of lithium carbonate and liquid water. If a crewmember exhales an average of 450L of carbon dioxide over the course of a day, what mass of lithium hydroxide would be needed to remove it?

First, write and balance the reaction. Then, state what you are converting to.

Equation: _____

I am converting from: _____ *to* _____

Finally, write out your conversion factor.

Stoichiometry Mixed Problems Answers

SA Adler



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