

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

## Assessment

## Quiz

## Section: Calculating Quantities in Reactions

In the space provided, write the letter of the term or phrase that best answers the question.

- \_\_\_\_\_ 1. The chemical equation  $\text{P}_4\text{O}_{10} + 6\text{H}_2\text{O} \rightarrow 4\text{H}_3\text{PO}_4$  can be correctly interpreted as
- 1 molecule of  $\text{P}_4\text{O}_{10} + 6$  molecules of  $\text{H}_2\text{O}$  form 4 molecules of  $\text{H}_3\text{PO}_4$ .
  - 1 mol  $\text{P}_4\text{O}_{10} + 6$  mol  $\text{H}_2\text{O}$  reacts to form 4 mol  $\text{H}_3\text{PO}_4$ .
  - 1 g  $\text{P}_4\text{O}_{10} + 6$  g  $\text{H}_2\text{O}$  produces 4 g  $\text{H}_3\text{PO}_4$ .
  - Both (a) and (b)
- \_\_\_\_\_ 2. Stoichiometry is the branch of chemistry that deals with the \_\_\_\_\_ between elements and compounds as reactants and products in a chemical reaction.
- bonding
  - energy transfers
  - numerical relationship
  - physical interactions
- \_\_\_\_\_ 3. How many mole ratios can be correctly obtained from the following chemical equation?
- $$\text{Na}_2\text{S} + \text{Cd}(\text{NO}_3)_2 \rightarrow 2\text{NaNO}_3 + \text{CdS}$$
- 1
  - 4
  - 12
  - 16
- \_\_\_\_\_ 4. Molar mass can be used to relate
- mass of reactant to moles of reactant.
  - moles of product to mass of product.
  - mass of product to moles of product.
  - All of the above
- \_\_\_\_\_ 5. In the chemical reaction described by the equation
- $$4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3(s),$$
- the mole ratio of iron(III) oxide to iron is
- 1:1.
  - 1:2.
  - 2:3.
  - 4:1.

# Stoichiometry Section Quiz

**Luann Marie Decker**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the name bar.

## Stoichiometry Section Quiz:

*Chemistry All-in-One For Dummies (+ Chapter Quizzes Online)* Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you ll need to pass your chemistry class with flying colors You ll find tons of practical examples and practice problems and you ll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter quizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or want to cram extra hard to ace that chem test

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