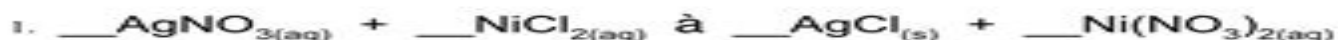


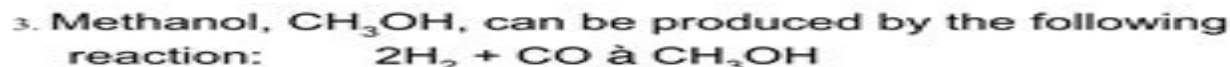
Stoichiometry & Limiting Reactant Review Sheet



- a) Determine the limiting reactant when 0.847 grams of silver nitrate is reacted with 0.650 grams of nickel(II) chloride. Show your calculations to justify your answer.
- b) What mass of precipitate will be produced?
- c) How many grams of the excess reactant will be left when the reaction stops?



- a) 80.0 grams of diiodine pentoxide reacts with 28.0 grams of carbon monoxide. Determine the limiting and excess reactants?
- b) What will be the mass of the iodine produced?
- c) What mass of the excess reactant will be left over?



- a) Calculate the mass of CH_3OH produced if 68.5 g of CO is reacted with 7.50 g of H_2 .
- b) Which substance would be the limiting reactant?
- c) How many grams of the excess reactant will be left over?



- a) Determine the limiting reactant when 6.76 grams of copper(II) sulfate is reacted with 9.82 grams of potassium iodide?
- b) What will be the mass of the iodine produced?
- c) What mass of the excess reactant will be left over?

Stoichiometry And Limiting Reactant Review

James E. Brady, John R. Holum



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