

Stoichiometry Worksheet # 1

- 1) Glucose is used as a source of energy by the human body. The overall reaction in the body is



Calculate the number of moles of oxygen needed to oxidize 12.5 g of $\text{C}_6\text{H}_{12}\text{O}_6$.

- 2) Ammonia is synthesized from hydrogen and nitrogen according to the following equation.



If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia (NH_3) can be produced?

- 3) Assume that in the decomposition of potassium chlorate, KClO_3 ,



80.5 moles of O_2 form. How many grams of potassium chlorite, the other product would be formed?

- 4) In a single displacement reaction,



9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

- 5) The compound "Cisplatin" $\text{PtCl}_2(\text{NH}_3)_2$, has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction



A. How many moles of "cisplatin" can be produced from 2.50 g K_2PtCl_4 ?

B. How many moles of NH_3 would be needed if 3.8 moles of K_2PtCl_4 were present?

Stoichiometry Chapter 12 Answer Key

SJ Ball



Stoichiometry Chapter 12 Answer Key:

This book delves into Stoichiometry Chapter 12 Answer Key. Stoichiometry Chapter 12 Answer Key is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Stoichiometry Chapter 12 Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Stoichiometry Chapter 12 Answer Key
- Chapter 2: Essential Elements of Stoichiometry Chapter 12 Answer Key
- Chapter 3: Stoichiometry Chapter 12 Answer Key in Everyday Life
- Chapter 4: Stoichiometry Chapter 12 Answer Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Stoichiometry Chapter 12 Answer Key. This chapter will explore what Stoichiometry Chapter 12 Answer Key is, why Stoichiometry Chapter 12 Answer Key is vital, and how to effectively learn about Stoichiometry Chapter 12 Answer Key.

3. In chapter 2, the author will delve into the foundational concepts of Stoichiometry Chapter 12 Answer Key. This chapter will elucidate the essential principles that must be understood to grasp Stoichiometry Chapter 12 Answer Key in its entirety.

4. In chapter 3, this book will examine the practical applications of Stoichiometry Chapter 12 Answer Key in daily life. The third chapter will showcase real-world examples of how Stoichiometry Chapter 12 Answer Key can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Stoichiometry Chapter 12 Answer Key in specific contexts. The fourth chapter will explore how Stoichiometry Chapter 12 Answer Key is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Stoichiometry Chapter 12 Answer Key. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Stoichiometry Chapter 12 Answer Key.

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