9.2 – Stoichiometry – Limiting Reagent
Practice: (u(12 +2NaNO3 -> Cu(NO3)2 +2NaC/
20 grams of copper (II) chloride react with 25 grams of sodium nitrate. Identify the limiting reagent. Need 20 grams of copper (II) chloride react with 25 grams of sodium nitrate. Identify the limiting reagent. Need 20 grams of copper (II) chloride react with 25 grams of sodium nitrate. Identify the limiting reagent. Need 20 grams of copper (II) chloride react with 25 grams of sodium nitrate. Identify the limiting reagent. Need 20 grams of copper (II) chloride react with 25 grams of sodium nitrate. Identify the limiting reagent. Need 137.4512 Graft No. 149 molloide x 2 mol No. 10 molloide x 2 molloide x
25g Notto, x 1 mol NaNO3 0.29+ mol Hatily x 1 mol with 0.147 no 1 84,994 g Hotto, 0.29+ mol Hatily well willy well
You need 0.298 mol NaNO3, but you have 0.294 mol NaNO,

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