Stoichiometry Problem Set A

Directions: Solve each of the following problems. Show work using the factor label method or the irf box. Include correct units and significant figures with each answer. Clearly indicate final answer. Be sure to start with a balanced equation

Part 1: Straight Stoichiometry Problems

Target Goal: To properly calculate the amount of one substance equal to a given amount of another substance within a chemical reaction using the factor-label method.

For the reaction HNO_{Neq} + Mg(OH)₁₀₀ → Mg(NO₂)_{3(eq)} + H₂O₍₀₎, how many grams of magnesium nitrate are produced from 8.00 mol of nitric acid (HNO₂)? A

MM=148.39

For the reaction 2 KClO₃₀₀ → 2 KCl₀₀ + 3 O₂₀₀, how many moles of potassium chlorate are required to produce 160. g of oxygen? A

MH = 32.09

- KCID3
- 3.) For the reaction $2H_{200} + O_{200} \rightarrow 2H_2O_{50}$, how many grams of water are produces from 5.22 liters of oxygen gas?

MH= 18.09

4.) In the reaction Na₍₀₎ + PH₂O₍₀₎ → NaOH₀₋₀ + H₂₍₀₎, how many grams of hydrogen are produced if 1.33 grams of sodium react completely?

MH= 23.09

5.) In the reaction $CO_{200} + H_2O_{00} \Rightarrow H_2O_{200}$, how many moles of carbonic acid can be produced from 97.2 g of CO_2 ?

MH 44.09

2.21 moles Heca

Did you achieve our target goal?

Target Goal: To properly calculate the amount of one substance equal to a given amount of another substance within a chemical reaction using the factor-label method. Circle your answer.

Feeling VERY confused

3 I think I am catching on

Doing great! Bring it on!

Stoichiometry Problem Set 1 Answers

Neil D. Jespersen, Pamela Kerrigan

Stoichiometry Problem Set 1 Answers:

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Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Advancing Assessment for Student Success Amy Driscoll, Swarup Wood, Dan Shapiro, Nelson Graff, 2023-07-03 This book is about student success and how to support and improve it It takes as its point of departure that we as faculty assessment directors student affairs professionals and staff reflect together in a purposeful and informed way about how our teaching curricula the co curriculum and assessment work in concert to support and improve student learning and success It also requires that we do so in collaboration with our colleagues and our students for the rich insights that we gain from them Conversational in style this book offers a wide variety of illustrations of how your peers are putting assessment into practice in ways that are meaningful to them and their institutions and that lead to improved student learning The authors provide rich guidance for activities ranging from everyday classroom teaching and assessment to using assessment to improve programs and entire institutions The authors envisage individual faculty at four year institutions and community colleges as their main audience whether those faculty are focused on their own classes or support their colleagues through leadership roles in assessment If you plan to remain focused on your own courses and students you will find that those sections of this book will help you better understand why and how assessment leaders do

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