

Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

- 18 mol O₂ 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?



$$\frac{12 \text{ mol KClO}_3}{2 \text{ mol KClO}_3} \quad | \quad \frac{3 \text{ mol O}_2}{2 \text{ mol KClO}_3}$$

- 21 mol O₂ 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?



$$\frac{14 \text{ mol KClO}_3}{2 \text{ mol KClO}_3} \quad | \quad \frac{3 \text{ mol O}_2}{2 \text{ mol KClO}_3}$$

- 45 mol H₂O 3. How many moles of water are produced from 22.5 moles of oxygen?



$$\frac{22.5 \text{ mol O}_2}{1 \text{ mol O}_2} \quad | \quad \frac{2 \text{ mol H}_2\text{O}}{1 \text{ mol O}_2}$$

- 20 mol HCl 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.



$$\frac{10 \text{ mol H}_2}{1 \text{ mol H}_2} \quad | \quad \frac{2 \text{ mol HCl}}{1 \text{ mol H}_2}$$

- 21 mol Cl₂ 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



$$\frac{14 \text{ mol FeCl}_3}{2 \text{ mol FeCl}_3} \quad | \quad \frac{3 \text{ mol Cl}_2}{2 \text{ mol FeCl}_3}$$

Stoichiometry Questions Answer Key

David Kirk



Stoichiometry Questions Answer Key:

Stoichiometry Questions Answer Key Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Stoichiometry Questions Answer Key**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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