

Chemistry--Unit 5: Stoichiometry  
Test Review

vocab

1. actual yield--the amount of product formed when a reaction is carried out in lab
2. excess reactant--a quantity of a reactant that is more than enough to react with a limiting reactant
3. limiting reactant--the reactant that determines the amount of product that can be formed in a reaction
4. percent yield--the ratio of the actual yield to the theoretical yield
5. stoichiometry--the calculation of quantities in chemical equations
6. theoretical yield--the calculated amount of product that should be formed during a reaction
7. titration--a method to determine unknown concentrations of solution from known concentrations of solutions using the neutralization reaction between acids and bases
8. equivalence point--when equal moles of acid and base are present during a titration and a chosen indicator changes color

Also know what is conserved and not conserved in a reaction

conserved	not conserved
grams, atoms	liters, moles

problems (stoichiometry, limiting reactant, percent yield, titrations)

problems to try

Given the balanced equation  $\text{Au}_2\text{S}_3(s) + 3\text{H}_2(g) \rightarrow 2\text{Au}(s) + 3\text{H}_2\text{S}(g)$

1. How many atoms of Au can be obtained from  $3.78 \times 10^{24}$  molecules of  $\text{Au}_2\text{S}_3$ ? ( $7.56 \times 10^{24}$  atoms Au)
2. How many moles of Au can be obtained from 12 moles of  $\text{H}_2$ ? ( $8 \text{ mol Au}$ )
3. How many grams of  $\text{H}_2\text{S}$  are formed if 3.75 moles of Au are formed during a reaction? ( $191.7 \text{ g H}_2\text{S}$ )
4. How many moles of  $\text{H}_2\text{S}$  can be formed from 36.54 g  $\text{Au}_2\text{S}_3$ ? ( $0.22 \text{ mol H}_2\text{S}$ )
5. If reactants and products are at the same temperature and pressure, how many liters of  $\text{H}_2$  are required to produce 6.97 L of  $\text{H}_2\text{S}$ ? ( $6.97 \text{ L H}_2$ )

# Stoichiometry Review Vocab

**Lei Shi**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Lei Shi'.

## **Stoichiometry Review Vocab:**

This book delves into Stoichiometry Review Vocab. Stoichiometry Review Vocab is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Stoichiometry Review Vocab, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Stoichiometry Review Vocab
- Chapter 2: Essential Elements of Stoichiometry Review Vocab
- Chapter 3: Stoichiometry Review Vocab in Everyday Life
- Chapter 4: Stoichiometry Review Vocab in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Stoichiometry Review Vocab. This chapter will explore what Stoichiometry Review Vocab is, why Stoichiometry Review Vocab is vital, and how to effectively learn about Stoichiometry Review Vocab.
3. In chapter 2, this book will delve into the foundational concepts of Stoichiometry Review Vocab. This chapter will elucidate the essential principles that must be understood to grasp Stoichiometry Review Vocab in its entirety.
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5. In chapter 4, the author will scrutinize the relevance of Stoichiometry Review Vocab in specific contexts. The fourth chapter will explore how Stoichiometry Review Vocab is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Stoichiometry Review Vocab. The final chapter will summarize the key points that have been discussed throughout the book.

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