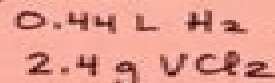


Name BETH "KEY" Period _____
Review Problems: Stoichiometry

1. Vanadium metal is prepared by the following unbalanced reaction:



How many liters of hydrogen are needed to produce 1.0 grams of vanadium metal at STP? How many grams of vanadium chloride are needed for this reaction?



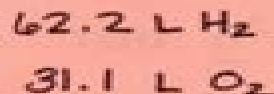
2. For the following unbalanced reaction:



how many grams of magnesium metal are needed to react completely with 50.0 g of phosphoric acid (H_3PO_4)? How many liters of hydrogen would be produced in this reaction at STP?



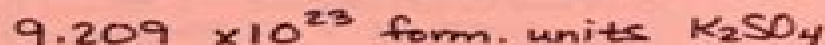
3. It is desired to prepare 50.0 grams of water by synthesis. How many liters of hydrogen must be used at STP? How many liters of oxygen at STP?



4. For the following unbalanced reaction between sulfuric acid and potassium hydroxide:



how many formula units of potassium sulfate can be produced if you have 150.0 grams of sulfuric acid to begin with and sufficient potassium hydroxide to completely react the sulfuric acid?



Stoichiometry Review Ws

S Marginson



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