

Practice Test Ch 3

Stoichiometry

Name: _____ Per: _____

- Remember, this is practice - Do NOT cheat yourself of finding out what you are capable of doing. Be sure you follow the testing conditions outlined below.
- DO NOT USE A CALCULATOR. AP Chem does not allow the use of a calculator for the MC part of the exam, so it is time to start practicing without one. You may use ONLY a periodic table.
- While you should practice working as fast as possible, it is more important at this point in the course, that you practice without a calculator, even if it slows you down. Look for the "easy math" – common factors and rough estimation – do not do "long division" to try to get exact values. Remember it is a MC test, use the answers
- Mark which questions you would like to "go over" when we get to school in September.

1. Balance the following equation:



The balanced equation shows that 1.00 mole of NH_3 requires _____ mole(s) of O_2 :

- a. 0.57
 - b. 1.25
 - c. 1.33
 - d. 1.75
 - e. 3.5
2. Write a balanced equation for the combustion of acetaldehyde, CH_3CHO .
- When properly balanced, the equation indicates that _____ mole(s) of O_2 are required for each mole of CH_3CHO .
- a. 1
 - b. 2
 - c. 2.5
 - d. 3
 - e. 5
3. What is the total mass of products formed when 16 grams of CH_4 is burned with excess oxygen?
- a. 32 g
 - b. 36 g
 - c. 44 g
 - d. 62 g
 - e. 80 g
4. Write a balanced equation for the combustion of propane, C_3H_8 .

When balanced, the equation indicates that _____ moles of O_2 are required for each mole of C_3H_8 .

- a. 1.5
- b. 3
- c. 3.5
- d. 5
- e. 8

5. Balance the following equation with the SMALLEST WHOLE NUMBER COEFFICIENTS possible. Select the number that is the
- sum
- of the coefficients in the balanced equation:



- a. 3
 - b. 5
 - c. 6
 - d. 7
 - e. 8
6. Calculate the mass of hydrogen formed when 27 g of aluminum reacts with excess hydrochloric acid according to the balanced equation below.
- $$2\text{Al} + 6\text{HCl} \rightarrow 2\text{AlCl}_3 + 3\text{H}_2$$
- a. 1.5 g
 - b. 2.0 g
 - c. 3.0 g
 - d. 6.0 g
 - e. 12 g
7. How many grams of nitric acid, HNO_3 , can be prepared from the reaction of 128 g of NO_2 with 54.0 g H_2O according to the equation below?
- $$3\text{NO}_2 + \text{H}_2\text{O} \rightarrow 2\text{HNO}_3 + \text{NO}$$
- a. 92
 - b. 108
 - c. 126
 - d. 189
 - e. 279
8. Which of the following statements is true?
- I. The molar mass of CaCO_3 is 100.1 g mol^{-1} .
 - II. 50 g of CaCO_3 contains 9×10^{23} oxygen atoms.
 - III. A 200 g sample of CaCO_3 contains 2 moles of CaCO_3 .

- a. I only
- b. II only
- c. III only
- d. I and III only
- e. I, II, and III

Stoichiometry Test Answers

Paul Foglino, Princeton Review (Firm)

Stoichiometry Test Answers:

MCAT Exam Prep Bill T Reese, 2024-05-13 What does it take to embark on the journey towards a career in medicine Dive into the world of aspiring medical professionals with our comprehensive guide designed to equip you with the knowledge skills and strategies needed to conquer the Medical College Admission Test MCAT and pave the way for success in medical school We unravel the mysteries of the MCAT through a series of engaging chapters that delve into the core subjects tested on the exam From biology and biochemistry to physics and organic chemistry each chapter offers a deep exploration of essential concepts accompanied by practice questions and detailed explanations to reinforce your understanding But mastering the MCAT is not just about memorizing facts and formulas it s about honing your critical thinking analytical reasoning and problem solving abilities That s why our book goes beyond mere content review to provide valuable insights into effective study strategies test taking techniques and mental preparation tips to help you perform at your best on exam day Whether you re a pre med student embarking on your MCAT journey or a seasoned test taker seeking to improve your scores this book is your trusted companion every step of the way Join us as we empower you to embrace the challenges seize the opportunities and embark on a transformative journey towards realizing your dreams of a career in medicine Prepare to embark on an enriching and empowering voyage one that will not only shape your academic future but also ignite your passion for healing compassion and making a meaningful difference in the world Are you ready to rise to the challenge and unlock your potential for greatness The journey begins here *Cracking the AP Chemistry Exam, 2013 Edition* Paul Foglino, Princeton Review, 2012-08-07 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests a subject review for all topics and sample questions and answers *AP Chemistry For Dummies* Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to

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Cracking the AP Chemistry Paul Foglino, Princeton Review (Firm), 2004 The fiercer the competition to get into college the more schools require that students prove themselves in other ways than SAT scores and grade point averages The more expensive college educations become the more students take advantage of the opportunity to test out off first year college courses Includes 2 sample tests with full explanations for all answers The Princeton Review s proven score raising skills and techniques Complete subject review of all the material likely to show up on the AP Chemistry exam

Cracking the AP Chemistry Exam, 2009 Edition Paul Foglino, 2009-01-01 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests

Intelligent Tutoring Systems Stefan Trausan-Matu, Kristy Boyer, Martha Crosby, Kitty Panourgia, 2014-05-24 This book constitutes the refereed proceedings of the 12th International Conference on Intelligent Tutoring Systems ITS 2014 held in Honolulu HI USA in June 2014 The 31 revised full papers 45 short papers and 27 posters presented were carefully viewed and selected from 177 submissions The specific theme of the ITS 2014 conference is Creating fertile soil for learning interactions Besides that the highly interdisciplinary ITS conferences bring together researchers in computer science learning sciences cognitive and educational psychology sociology cognitive science artificial intelligence machine learning and linguistics The papers are organized in topical sections on affect multimodality and metacognition collaborative learning data mining and student behavior dialogue and discourse generating hints scaffolds and questions game based learning and simulation graphical representations and learning student strategies and problem solving scaling ITS and assessment

Resources in Education, 1995

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Roger Slade, Maureen Slade, 2012-12-14 Cambridge Checkpoints VCE are updated regularly to provide you with the most up to date exam preparation available **Holt Chemistry** Ralph Thomas Myers, 2004

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