



Vocabulary: Stoichiometry



Vocabulary

- Avogadro's number – the number of atoms or molecules in a mole of a substance.
 - Avogadro's number is equal to 6.0221415×10^{23} .
- Balanced equation – an equation that has equal numbers of atoms on each side of the equation.
 - The equation $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ is *unbalanced* because there are two oxygen atoms on the left side and only one on the right.
 - The equation $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ is balanced because there are four hydrogen atoms and two oxygen atoms on each side of the equation.

- Cancel – to cross out, remove, delete, or invalidate something.
 - In an algebraic expression, identical terms that are found in the numerator and denominator can be cancelled, such as the unit “seconds” in this equation:

$$120 \text{ ~~seconds~~} \cdot \frac{1 \text{ minute}}{60 \text{ ~~seconds~~}} = 2 \text{ minutes}$$

- Coefficient – a number that multiplies a term in an equation.
 - In a chemical equation, the coefficients indicate the number of each type of molecule. For example, $6\text{H}_2\text{O}$ means that there are six water molecules.
- Conversion factor – a ratio or fraction equivalent to one that is used to multiply or divide a quantity when converting from one unit to another.
 - For example, the conversion factor for moles of CO_2 to liters of CO_2 is:

$$\frac{1 \text{ mol CO}_2}{22.41 \text{ L CO}_2}$$

This conversion factor is equivalent to one because one mole of carbon dioxide has a volume of 22.41 liters.

- Dimensional analysis – a technique in which comparison of units on both sides of an equation is used to solve problems.
- Formula mass – the mass of one unit of a compound, as measured in unified mass units (u). For example, to find the formula mass of NaCl
 - For example, the formula mass of NaCl is found by adding the average atomic masses of sodium (Na) and chlorine (Cl): $22.99 \text{ u} + 35.45 \text{ u} = 58.44 \text{ u}$.



Stoichiometry Vocabulary Answer

SA Adler



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Table of Contents **Stoichiometry Vocabulary Answer**

1. Understanding the eBook **Stoichiometry Vocabulary Answer**
 - The Rise of Digital Reading **Stoichiometry Vocabulary Answer**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Stoichiometry Vocabulary Answer**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a **Stoichiometry Vocabulary Answer**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Stoichiometry Vocabulary Answer**
 - Personalized Recommendations
 - **Stoichiometry Vocabulary Answer** User Reviews and Ratings
 - **Stoichiometry Vocabulary Answer** and Bestseller Lists

5. Accessing Stoichiometry Vocabulary Answer Free and Paid eBooks
 - Stoichiometry Vocabulary Answer Public Domain eBooks
 - Stoichiometry Vocabulary Answer eBook Subscription Services
 - Stoichiometry Vocabulary Answer Budget-Friendly Options
6. Navigating Stoichiometry Vocabulary Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Stoichiometry Vocabulary Answer Compatibility with Devices
 - Stoichiometry Vocabulary Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Stoichiometry Vocabulary Answer
 - Highlighting and Note-Taking Stoichiometry Vocabulary Answer
 - Interactive Elements Stoichiometry Vocabulary Answer
8. Staying Engaged with Stoichiometry Vocabulary Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Stoichiometry Vocabulary Answer
9. Balancing eBooks and Physical Books Stoichiometry Vocabulary Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Stoichiometry Vocabulary Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Stoichiometry Vocabulary Answer
 - Setting Reading Goals Stoichiometry Vocabulary Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Stoichiometry Vocabulary Answer
 - Fact-Checking eBook Content of Stoichiometry Vocabulary Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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