

# Vocabulary: Stoichiometry



#### ∇ocabulary

- Avogadro's number the number of atoms or molecules in a mole of a substance.
  - Avogadro's number is equal to 6.0221415 × 10<sup>23</sup>.
- Balanced equation an equation that has equal numbers of atoms on each side of the equation.
  - The equation H₂ + O₂ → H₂O is unbalanced because there are two oxygen atoms. on the left side and only one on the right.
  - The equation 2H₂ + O₂ → 2H₂O is balanced because there are four hydrogen. atoms and two oxygen atoms on each side of the equation.
- Cancel to cross out, remove, delete, or invalidate something.
  - In an algebraic expression, identical terms that are found in the numerator and denominator can be cancelled, such as the unit "seconds" in this equation:

120 seconds - 1 minute = 2 minutes 80 seconds

- Coefficient a number that multiplies a term in an equation.
  - In a chemical equation, the coefficients indicate the number of each type of molecule. For example, 6H<sub>2</sub>O means that there are six water molecules.
- Conversion factor a ratio or fraction equivalent to one that is used to multiply or divide a quantity when converting from one unit to another.
  - For example, the conversion factor for moles of CO<sub>2</sub> to liters of CO<sub>2</sub> is:

This conversion factor is equivalent to one because one mole of carbon dioxide has a volume of 22.41 liters.

- Dimensional analysis a technique in which comparison of units on both sides of an equation is used to solve problems.
- Formula mass the mass of one unit of a compound, as measured in unified mass units (u). For example, to find the formula mass of NaCl
  - For example, the formula mass of NaCl is found by adding the average atomic masses of sodium (Na) and chlorine (CI): 22.99 u + 35.45 u = 58.44 u.



# **Stoichiometry Vocabulary Answers**

**Ying-Ying Zheng** 

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