### Stoichiometry Calculation Practice Worksheet

 Calculate the number of moles of NaOH that are needed to react with 500.0 g of H<sub>2</sub>SO<sub>4</sub> according to the following equation:

$$H_2SO_4 + 2 NaOH \rightarrow Na_2SO_4 + 2 H_2O$$

 Calculate the mass of NH<sub>3</sub> that can be produced from the reaction of 125 g of NCl<sub>3</sub> according to the following equation:

$$NCl_3 + 3 H_2O \rightarrow NH_3 + 3 HOC1$$

 What mass of CO<sub>2</sub> can be produced from the reaction of 25.0 g of C<sub>3</sub>H<sub>8</sub> with 75.0 g of O<sub>2</sub> according to the following equation:

$$C_3H_8$$
 + 5  $O_2$   $\rightarrow$  3  $CO_2$  + 4  $H_2O$ 

# **Stoichiometry 13 Practice Problems**

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