



Name: Chelsey Williams Date: 3/31/22

Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, formula mass, molar mass, mole, molecular mass, stoichiometry

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A 250 mL glass of orange juice contains 22 grams of sugar. How much sugar is in a two-liter (2,000 mL) bottle of orange juice? 176 grams
2. It requires two sticks of butter to make a batch of 20 cookies. How much butter will it take to make 150 cookies? 15 sticks of butter

Gizmo Warm-up

Just as a cook follows a recipe to decide how much of each ingredient to add, a chemist uses **stoichiometry** to determine the amounts of substances involved in chemical reactions. The Stoichiometry Gizmo allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction.

To begin, check that this equation is shown:



1. Look at the **coefficients** (such as the "3" in 3CO) in front of each substance in the equation. The coefficients tell you how many molecules or atoms take part in a chemical reaction. In the spaces below, list the number of each molecule or atom in the equation:

Fe_2O_3 1

CO 3

Fe 2

CO_2 3

2. In a **balanced equation**, the same number of each kind of atom is shown on each side of the equation. Calculate the number of iron (Fe), oxygen (O), and carbon atoms (C).

New equation



New question

How many moles of carbon monoxide (CO) are required to react completely with 1.75 moles of iron (III) oxide (Fe_2O_3)?

Problems solved: 0



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Sabine Zange



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Chemical Mathematics and Biostatistics Mr. Rohit Manglik, 2024-02-24 Applies mathematical techniques and statistical methods in chemistry and biology for data interpretation and research analysis **Chemistry** James E. Brady, John R. Holum, 1995-12-29 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving *Ncert Objective Textbook- Chemistry* Dr. Manish Rannjan (IAS), 2021-01-19 NCERT Objective Textbook Chemistry by Dr Manish Rannjan IAS NCERT Objective Textbook Chemistry by Dr Manish Rannjan IAS is a comprehensive guidebook designed to help students understand and excel in chemistry as per the NCERT curriculum This book covers the essential topics of chemistry including atomic structure chemical bonding states of matter organic chemistry and more With its objective based approach practice questions and detailed explanations this book aids students in developing a strong foundation in chemistry and prepares them for various competitive exams Key Aspects of the Book NCERT Objective Textbook Chemistry NCERT Curriculum Coverage The book aligns with the NCERT curriculum ensuring comprehensive coverage of the chemistry topics prescribed in the syllabus It follows a structured and systematic approach to present the concepts making it easier for students to grasp and retain the information Objective Based Questions The book includes a wide range of objective based questions including multiple choice fill in the blanks and matching type questions These questions are designed to test students understanding of the concepts and help them practice for competitive exams where objective type questions are prominent Detailed Explanations and Solutions Each question in the book is accompanied by detailed explanations and solutions This helps students understand the underlying concepts and the step by step approach to solving problems The detailed explanations aid in clarifying doubts and reinforcing the understanding of the subject Dr Manish Rannjan IAS the author of NCERT Objective Textbook Chemistry is an esteemed educator and accomplished civil servant With his extensive knowledge and experience Dr Rannjan aims to simplify the study of chemistry for students and provide them with a comprehensive resource that aligns with the NCERT curriculum His book is a valuable tool for students preparing for various competitive exams enabling them to develop a solid foundation in chemistry and achieve academic success *Study Guide, Chemical Principles, Fifth Edition, Zumdahl* Paul B. Kelter, Steven S. Zumdahl, 2004-04 *Journal of the Society of Chemical Industry* Society of Chemical Industry (Great Britain), 1908 **Foundations of College Chemistry, Alternate** Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis **Journal of the Society of Chemical Industry**

,1908 How to Become a Successful Electrician Thomas O'Connor Sloane,1906 *Chemistry: The Central Science* Theodore L. Brown,H. Eugene LeMay Jr.,Bruce E. Bursten,Catherine Murphy,Patrick Woodward,Steven Langford,Daluis Sagatys,Adrian George,2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **The Quest for Food** Harald Brüssow,2007-05-11 This book explores the links between food and human cultural and physical evolution Each chapter begins by summarizing the basic knowledge in the field discusses recent research results and confirms or challenges established concepts inviting new insight and provoking new questions This book catalyzes discussion between scientists working on one side in food science and on the other side in biological and biomedical research Molecular Aspects of Botulinum Neurotoxin Keith A. Foster,2014-07-08 Currently there are tremendous advances being made in understanding the basic science of both the structure and function of botulinum neurotoxins This knowledge is opening up opportunities in regard to both therapeutic uses and treatment and protection options for civil and bio defense applications This volume fully evaluates the status of neurotoxin research and exploitation with a focus on clinical application The book is a multi authored collection of chapters written by the leading authorities responsible for the current scientific and clinical research that is advancing the understanding and exploitation of the neurotoxins and is both up to date and authoritative **Choice** ,2002 **Metric, Myth & Quasicrystals** Antony J. Bourdillon,2012-08 Metric Myth the series is not properly Fibonacci simplified indexation of the pattern is used to simulate a structure due to a single aligned edge sharing unit cell that is consistent with all data Since it is unlikely that the sharp diffraction pattern is due to unmeasured poly polyhedra does the International Union of Crystallography have to redefine crystals yet again In modern physics the metric relates the covariant components of invariant vectors with corresponding contravariant components In crystallography it relates dimensions in momentum space to atomic locations in real space In quasicrystals the pattern in momentum space is logarithmic Theory and simulation show why this has to be Consequences follow In particular we show not only where the atoms are but also why they are there A debate is reported so that the reader will be encouraged to make his own mind When logarithmic periodicity is discovered and explained in one branch of physics it should be expected in others **Cumulated Index Medicus** ,1974 *Chemical Abstracts* ,1990 The Publishers' Trade List Annual ,1974 **Sci-tech Book Profiles** ,1965 Includes title page table of contents list of contributors preface

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