Name	Date

#### Chapter 12 – Stoichiometry Study Guide

The following example problems exhibit the types of calculations you will be expected to perform on tomorrow's test.

Describe a reaction in terms of particles

Ex.) Describe the following reaction in terms of molecules

Note: sometimes you will need to include other descriptors such as atoms and formula units\*

Describe a reaction in terms of moles
 Ex.) Describe the following reaction in terms of moles

How do you determine the mass of the products for a balanced reaction?
 Ex.) What is the mass of the product formed in the following equation?

$$H_2 + CI_2 \rightarrow 2HCI$$

 How do you determine the volume of the reactants or products of a balanced reaction?

Ex.) What is the volume of the products and the reactants of the following equation?

$$2H_2 + O_2 \rightarrow 2H_2O$$

 How do you show how a balanced reaction follows the Law of Conservation of Mass by showing that the mass on both sides of a reaction are equal?

Ex.) Show how the mass of the reactants equals the mass of the products for the following equation.

# **Stoichiometry Chapter 1study Guide Answer Key**

**Davide Dionisi** 

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