

## Chapter Test B

## A. Matching

Match each item in Column A with the correct description in Column B. Write the letter of the correct description in the blank.

## Column A

1. An unknown mass dissolved in a known volume of solvent for solution density, a molar.
2. An unknown volume of gas produced from a known mass of a solid reactant for the molar mass.
3. The relationship of quantities in chemical equations.
4. The ratio of the actual yield to the theoretical yield, expressed as a percent.
5. The unknown mass of a product or reactant in a chemical equation for a molar mass.
6. An unknown volume of a gas produced from a known mass of a solid reactant for the molar mass.

## Column B

- a. percent yield
- b. limiting reagent
- c. theoretical yield
- d. stoichiometry
- e. molar mass
- f. molar volume
- g. molar weight

## B. Multiple Choice

Choose the best answer for each question from the list.

1. A 10.0-g sample of a mixture of  $\text{NaCl}$  and  $\text{KCl}$  contains 4.0 g of  $\text{NaCl}$ . What is the mass percent of  $\text{KCl}$  in the mixture?  
a. 20.0%  
b. 40.0%  
c. 60.0%  
d. 80.0%
2. A 10.0-g sample of a mixture of  $\text{NaCl}$  and  $\text{KCl}$  contains 4.0 g of  $\text{NaCl}$ . What is the mass percent of  $\text{NaCl}$  in the mixture?  
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