

4. a.
$$\frac{10 \text{ g Al} \mid 1 \text{ mol Al} \mid 3 \text{ mol Cu} \mid 63.55 \text{ g Cu}}{26.98 \text{ g Al} \mid 2 \text{ mol Al} \mid 1 \text{ mol Cu}} = 35.33 \text{ g Cu}$$

$$\frac{20 \text{ g CuCl}_2 \mid 1 \text{ mol CuCl}_2 \mid 3 \text{ mol Cu} \mid 63.55 \text{ g Cu}}{134.45 \text{ g CuCl}_2 \mid 3 \text{ mol CuCl}_2 \mid 1 \text{ mol Cu}} = \boxed{9.45 \text{ g Cu}}$$

b.
$$\frac{30 \text{ g Al} \mid 1 \text{ mol Al} \mid 2 \text{ mol AlCl}_3 \mid 133.33 \text{ g AlCl}_3}{26.98 \text{ g Al} \mid 2 \text{ mol Al} \mid 1 \text{ mol AlCl}_3} = 148.25 \text{ g AlCl}_3$$

$$\frac{100 \text{ g CuCl}_2 \mid 1 \text{ mol CuCl}_2 \mid 2 \text{ mol AlCl}_3 \mid 133.33 \text{ g AlCl}_3}{134.45 \text{ g CuCl}_2 \mid 3 \text{ mol CuCl}_2 \mid 1 \text{ mol AlCl}_3} = \boxed{66.11 \text{ g AlCl}_3}$$

c.
$$\frac{5 \text{ g Al} \mid 1 \text{ mol Al} \mid 3 \text{ mol Cu} \mid 63.55 \text{ g Cu}}{26.98 \text{ g Al} \mid 2 \text{ mol Al} \mid 1 \text{ mol Cu}} = 17.67 \text{ g Cu}$$

~~100 g / 17.67 g~~ $10 \text{ g} / 17.67 \text{ g} \times 100 = \boxed{56.6\%}$

d.
$$\frac{10 \text{ g CuCl}_2 \mid 1 \text{ mol CuCl}_2 \mid 2 \text{ mol AlCl}_3 \mid 133.33 \text{ g AlCl}_3}{134.45 \text{ g CuCl}_2 \mid 3 \text{ mol CuCl}_2 \mid 1 \text{ mol AlCl}_3} = 6.61 \text{ g AlCl}_3$$

~~100 g / 6.61 g~~ $5 \text{ g} / 6.61 \text{ g} \times 100 = \boxed{75.6\%}$

- 5.
- a. TRUE
 - b. FALSE
 - c. $\text{AgNO}_3 \Rightarrow 169.91 \text{ g/mol}$
 - d. 6.02×10^{23} particles = 1 mole of particles

Stoichiometry Study Guide Answers

Arshad Iqbal



Stoichiometry Study Guide Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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