

1800cm³ | 22.4L | 4Cl₂

CH₄

Syring carbonate, NaHCO₃, combines with HCl according to the equation



of HCl should be present to combine totally with 0.250 moles of

mol NaHCO ₃		1 HCl		36.5g	= 9.13g HCl
		1 NaHCO ₃		1 mol	

grams of NaCl can be produced if you add 5.00 g of NaOH 5.00 grams



5.0g NaOH		1 mol NaOH		1 NaCl		58.5g	= 11.7g
		40g		1 NaOH		1 mol	

5.0g HCl		1 mol HCl		1 NaCl		58.5g	= 8.01g
		36.5g		1 HCl		1 mol	

grams of each product will be produced from 531 grams of C₅H₁₂



C ₅ H ₁₂		1 mol C ₅ H ₁₂		5 CO ₂		22.4L	= 820
		72.1g		1 C ₅ H ₁₂		1 mol	

Stoichiometry Unit Test Answers 7

Luann Marie Decker

A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the author's name.

Stoichiometry Unit Test Answers 7:

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