Stoichiometry Lab: Iron and Copper Sulfate Sample data

Data and observations :

mass of copper (Π) sulfate pentahydrate	3.27	9
mass of iron:	0.50	9
mass of filter paper	1.28	9
mass of filter paper and dry copper:	1.84	9
mass of dry copper	0.56	9

Observations of reaction:

When the iron was added a precipitate began to form. The precipitate was reddish in color. The solution remained a blue color but was somewhat lighter in color.

Calculations:

1&2.The single replacement equation will be: Fe + CuSO₄ → FeSO₄ + Cu

Fe and CuSO₄ are the reactants and FeSO₄ and Cu are the products.

4. The moles of copper (II) sulfate pentahydrate used were 0.013 mol.

$$3.27 \text{ g CuSO}_4 \cdot 5H_2O \times \frac{1 \text{ mol CuSO}_4 \cdot 5H_2O}{249.47 \text{ g CuSO}_4 \cdot 5H_2O} = 0.013 \text{ mol CuSO}_4 \cdot 5H_2O$$

5. The moles of iron used were 0.009 mol Fe.

$$0.50 \text{ g Fe} \times \frac{1 \text{ mol Fe}}{55.85 \text{ g Fe}} = 0.009 \text{ mol Fe}$$

The moles of copper produced were 0.009 mol Cu.

$$0.56 \text{ g Cu} \times \frac{1 \text{ mol Cu}}{63.55 \text{ g Cu}} = 0.009 \text{ mol Cu}$$

Stoichiometry Lab Copper Sulfate And Iron Report

Didier Musso

Stoichiometry Lab Copper Sulfate And Iron Report:

Laboratory Manual to Accompany Chemistry, [by] Stanley R. Radel, Marjorie H. Navidi Arthur D. Baker, 1990 Scientific and Technical Aerospace Reports ,1981 **Chemistry** Frank Jenkins, 1992 U.S. Government Research Reports, 1961 Fossil Energy Update ,1984 **Energy Research Abstracts** ,1987 **Toxicology Research Projects Directory**, 1976-03 An indexed directory of current research project abstracts in toxicology and related fields **Government Reports Announcements & Index** ,1995-05 Research and Development Abstracts of the USAEC U.S. Atomic Energy Commission, 1963-10 Research and Development Abstracts of the USAEC. U.S. Atomic Energy Commission. Division of Technical Information. 1962 The Environmental Geochemistry of Mineral Deposits Geoffrey S. Plumlee, Mark J. Logsdon, Lorraine H. Filipek, 1999 Bibliography of Scientific and Industrial Reports, 1968 Toxicology of Molluscicides Gerald Webbe, 1987 The past decades have seen the implementation of numerous programmes for the eradication of molluscs due mainly to their involvement as vectors of human parasitic diseases These programmes have involved the use of vast quantities of molluscicides and have resulted in many unwanted side effects This volume of the International Encyclopedia of Pharmacology and Therapeutics brings together reviews on the major molluscicides in current use Each review outlines all the physicochemical toxicological and pharmacological properties as well as detailing the effects on all the other major organisms that they are likely to come into contact with In bringing together all this information in one volume this book will enable the use of molluscicides to be better controlled and far more effective whilst preventing the serious ecological damage often associated with such programmes Part 1. Proterozoic Iron and Zinc Deposits of the Adirondack Mountains of the New York and the New Jersey Highlands John F. Slack, Robert R. Seal, Jane Marie Plating and Surface Finishing, 1991 Hammarstrom, 2001 Cumulated Index Medicus ,1977 Selected Water Resources Abstracts ,1982-10 Nuclear Science Abstracts, 1966 Physics Briefs ,1994 Metals Abstracts ,1999

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