

# LIMITING REACTANT STOICHIOMETRY WORKSHEET

Name _____	Section _____	Date _____
<b>Limiting Reactants</b>		
<b>Directions:</b> Determine the limiting reactant in each problem.		
1) In $2\text{NaOH} + \text{MgCl}_2 \rightarrow \text{Mg(OH)}_2 + 2\text{NaCl}$ , 5.818 grams of NaOH react with 2.712 grams of $\text{MgCl}_2$ .		
2) 0.0718 grams of HCl react with 0.0002 grams of $\text{Ca(OH)}_2$ in the reaction $2\text{HCl} + \text{Ca(OH)}_2 \rightarrow 2\text{H}_2\text{O} + \text{CaCl}_2$ .		
3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2008 grams of Hydrogen.		

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3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2008 grams of Hydrogen.		
4) In the reaction $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$ , 10.000 grams of $\text{O}_2$ react with 0.908 grams of $\text{C}_2\text{H}_2$ .		
5) In the reaction $2\text{Al(OH)}_3 + 3\text{H}_2\text{CO}_3 \rightarrow \text{Al}_2(\text{CO}_3)_3 + 6\text{H}_2\text{O}$ , 1.18 grams of $\text{Al(OH)}_3$ and 1.012 grams of $\text{H}_2\text{CO}_3$ combine.		
6) 0.838 grams of sodium and 0.7021 grams of sodium nitrate react in the following process: $10\text{Na} + 2\text{NaNO}_3 \rightarrow 6\text{Na}_2\text{O} + \text{N}_2$ .		
7) 0.0700 grams of $\text{AlBr}_3$ react with 1.201 grams of $\text{H}_2\text{SO}_4$ to produce products in the following reaction: $2\text{AlBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 6\text{HBr}$ .		
8) When $\text{K}_3\text{PO}_4$ and NaOH react in $\text{K}_3\text{PO}_4 + 3\text{NaOH} \rightarrow 3\text{KOH} + \text{Na}_3\text{PO}_4$ , 5.41 grams of NaOH react with 3.88 grams $\text{K}_3\text{PO}_4$ .		
9) In $\text{Pt}_2(\text{NO})_2 + \text{H}_2\text{S} \rightarrow \text{PtS} + \text{HNO}_3$ , 10.02 grams of $\text{Pt}_2(\text{NO})_2$ react with 7.98 grams of $\text{H}_2\text{S}$ .		
10) 0.4000 grams of $\text{H}_2\text{SO}_4$ combine with 0.2480 grams of $\text{Ba(OH)}_2$ in the reaction $2\text{H}_2\text{SO}_4 + \text{Ba(OH)}_2 \rightarrow \text{Ba(SO}_4)_2 + 4\text{H}_2\text{O}$ .		

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A CHEMISTRY WORKSHEET

10 PROBLEMS

WORKED ANSWER KEY

# Stoichiometry Section 12 3 Limiting Reactants Answer Key

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Jr., Bruce E. Bursten, Catherine  
Murphy, Patrick Woodward, Steven  
Langford, Dalius Sagatys, Adrian  
George**

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*Teaching Science for Understanding* James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities

*Instructor's Resource Manual* Silberberg, 1999-07

*Chemistry* John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

**Prentice Hall Chemistry**, 2000

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## **Table of Contents Stoichiometry Section 12 3 Limiting Reactants Answer Key**

1. Understanding the eBook Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - The Rise of Digital Reading Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Personalized Recommendations
  - Stoichiometry Section 12 3 Limiting Reactants Answer Key User Reviews and Ratings

- Stoichiometry Section 12 3 Limiting Reactants Answer Key and Bestseller Lists
- 5. Accessing Stoichiometry Section 12 3 Limiting Reactants Answer Key Free and Paid eBooks
  - Stoichiometry Section 12 3 Limiting Reactants Answer Key Public Domain eBooks
  - Stoichiometry Section 12 3 Limiting Reactants Answer Key eBook Subscription Services
  - Stoichiometry Section 12 3 Limiting Reactants Answer Key Budget-Friendly Options
- 6. Navigating Stoichiometry Section 12 3 Limiting Reactants Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Stoichiometry Section 12 3 Limiting Reactants Answer Key Compatibility with Devices
  - Stoichiometry Section 12 3 Limiting Reactants Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Highlighting and Note-Taking Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Interactive Elements Stoichiometry Section 12 3 Limiting Reactants Answer Key
- 8. Staying Engaged with Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Stoichiometry Section 12 3 Limiting Reactants Answer Key
- 9. Balancing eBooks and Physical Books Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Stoichiometry Section 12 3 Limiting Reactants Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Setting Reading Goals Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Fact-Checking eBook Content of Stoichiometry Section 12 3 Limiting Reactants Answer Key
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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