## Acid, Base or Salt

Classify each of the following compounds as an acid, base or salt. Then, indicate whether each acid and base is strong or weak.

1.	HNO <sub>3</sub>		
2.	NaOH		The state of the s
з.	NaNO <sub>3</sub>	- <del></del>	
4.	HCI	3 ·	
5.	KCI		
6.	Ba(OH) <sub>2</sub>		
7.	KOH		
8.	H <sub>2</sub> S		
9	AI(NO <sub>3</sub> ) <sub>3</sub>		
10.	H <sub>2</sub> SO <sub>4</sub>		
11.	CaCl <sub>2</sub>		
12.	H <sub>3</sub> PO <sub>4</sub>		
13.	Na <sub>2</sub> SO <sub>4</sub>	304	
14.	Mg(OH) <sub>2</sub>	45	
15.	H <sub>2</sub> CO <sub>3</sub>	***	
16.	NH₄OH	400	-
17.	NH <sub>4</sub> CI	and the second s	
18.	HBr		
19.	FeBr <sub>3</sub>		
20.	HF		

# **Reading Essentials Acids Bases And Salts Answer Key**

**WJ Hussar** 

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