ACS Gen Chem 1 Exam

Ion - A species of an element in which the number of electrons does not equal the number of protons.

Isotope - Each isotope of an atom has a different number of neutrons.

Isoelectric - Two atoms with the same charge.

Isotopic - Two atoms with the same number of neutrons.

Transition Metal Orbitals - The 4s orbital is filled before the 3d orbital, but listed after.

Paramagnetic - Atoms that possess a permanent magnetic charge due to the presence of unpaired electrons.

Electropositive - Tending to lose electrons and form positive ions.

Tetrahedral Bond Angle - 109.5°

Resonance Structures - Two equally correct arrangements of one molecule's electrons; generally the molecule exists as an average of the two.

Polar Molecule - A molecule that is mostly positive on one side and mostly negative on the other.

sp³ Orbital - An orbital composed of one s orbital and three p orbitals; can hold eight total electrons. Tetrahedral.

Trigonal Planar - The way that three objects will arrange themselves around a central atom. Non-polar.

Free Radicals - Compounds with an odd number of electrons in their Lewis Structure. Eg. NO₂

Incomplete Octets - Occur in elements that can form stable bonds with less than eight electrons. Eg. BF₃

Expanded Octets - Occur in the third row of the periodic table and beyond in elements that can form stable bonds with more than eight electrons. Eg. XeF₂

Hydrogen Bonding - When hydrogen is attracted to electronegative atoms.

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recognition elements should initiate or improve these performance factors. The technique discussed primarily applies to transducers involved in converting a biochemical signal to optical or electrical signals Over the years the emergence of novel materials and techniques has drastically improved the performance of these bioanalytical systems enabling them to expand their analytical horizon These advanced materials and techniques are central to modern bioanalytical and biosensor research Advanced Materials and Techniques for Biosensors and Bioanalytical Applications provides a comprehensive review of the subject including a knowledge platform for both academics and researchers Considering biosensors as a central theme to this book an outline on this subject with background principles has been included with a scope of extending the utility of the book to coursework in graduate and postgraduate schools Features Basic principles on different classes of biosensors recent advances and applications Smart materials for biosensors and other rapid portable detection devices Metal nanoparticles and nanocrystals for analytical applications Carbon based nanoparticles and quantum dots for sensing applications Nanozymes as potential catalysts for sensing applications Bioelectrochemiluminescence and photoelectrochemical based biosensors Paper electronics and paper based biosensors Microbial biosensors artificial intelligence genetic engineering and synthetic biology Biofuel cells as a signal transduction platform FET based biosensors including ISFET and BioFET This book serves as a reference for scientific investigators and a textbook for a graduate level course in biosensors and advanced bioanalytical Code of Federal Regulations, Title 21, Food and Drugs, PT. 170-199, Revised as of April 1, 2009, 2009-07 techniques Energy Research Abstracts, 1985

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