

## The Advanced Placement Examination in Chemistry

Part I – Multiple Choice Questions

Part II - Free Response Questions

Selected Questions from 1970 to 2010

## Stoichiometry

## Part I

1984

2. Which of the following forms a compound having the formula  $\text{KNO}_3$ ?
- (A) F  
(B) S  
(C) Mg  
(D) Ar  
(E) Mn
32. The net ionic equation for the reaction between silver carbonate and hydrochloric acid is
- (A)  $\text{Ag}_2\text{CO}_3(s) + 2 \text{H}^+ + 2 \text{Cl}^- \rightarrow 2 \text{AgCl}(s) + \text{H}_2\text{O} + \text{CO}_2(g)$   
(B)  $2 \text{Ag}^+ + \text{CO}_3^{2-} + 2 \text{H}^+ + 2 \text{Cl}^- \rightarrow 2 \text{AgCl}(s) + \text{H}_2\text{O} + \text{CO}_2(g)$   
(C)  $\text{CO}_3^{2-} + 2 \text{H}^+ \rightarrow \text{H}_2\text{O} + \text{CO}_2(g)$   
(D)  $\text{Ag}^+ + \text{Cl}^- \rightarrow \text{AgCl}(s)$   
(E)  $\text{Ag}_2\text{CO}_3(s) + 2 \text{H}^+ \rightarrow 2 \text{Ag}^+ + \text{H}_2\text{CO}_3$
44. What number of moles of  $\text{O}_2$  is needed to produce 14.2 grams of  $\text{P}_2\text{O}_{10}$  from P? (Molecular weight  $\text{P}_2\text{O}_{10} = 284$ )
- (A) 0.0500 mole  
(B) 0.0625 mole  
(C) 0.125 mole  
(D) 0.250 mole  
(E) 0.500 mole
45. The alkenes are compounds of carbon and hydrogen with the general formula  $\text{C}_n\text{H}_{2n}$ . If 0.561 gram of any alkene is burned in excess oxygen, what number of moles of  $\text{H}_2\text{O}$  is formed?
- (A) 0.0400 mole  
(B) 0.0600 mole  
(C) 0.0800 mole  
(D) 0.400 mole  
(E) 0.800 mole
52.  $3 \text{Ag}(s) + 4 \text{HNO}_3 \rightarrow 3 \text{AgNO}_3 + \text{NO}(g) + 2 \text{H}_2\text{O}$   
The reaction of silver metal and dilute nitric acid proceeds according to the equation above. If 0.10 mole of powdered silver is added to 10. milliliters of 6.0-molar nitric acid, the number of moles of NO gas that can be formed is
- (A) 0.015 mole  
(B) 0.020 mole  
(C) 0.030 mole  
(D) 0.045 mole  
(E) 0.090 mole



# Old Multiple Choice Questions For Stoichiometry

**Tobias Bleicker**





## **Old Multiple Choice Questions For Stoichiometry:**



## Old Multiple Choice Questions For Stoichiometry Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Old Multiple Choice Questions For Stoichiometry**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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