

THE URINARY SYSTEM

Metabolism of nutrients by body cells produces various wastes such as carbon dioxide and nitrogenous wastes (creatinine, urea, and ammonia), as well as imbalances of water and essential ions. The metabolic wastes and excesses must be eliminated from the body. Essential substances are retained to ensure proper body functioning.

Although several organ systems are involved in excretory processes, the urinary system bears the primary responsibility for removing nitrogenous wastes from the blood. In addition to this purely excretory function, the kidneys maintain the electrolyte, acid-base, and fluid balances of the blood. Thus, kidneys are major homeostatic organs of the body. Malfunction of the kidneys leads to a failure of homeostasis, resulting (unless corrected) in death.

Student activities in this chapter are concerned with identification of urinary system structures and with examining urine composition and physiological processes involved in urine formation. It also focuses on the composition of the body's fluid compartments and the water, electrolyte, and acid-base balance of these compartments.

1. Complete the following statements by inserting your answers in the answer blanks.

- The kidney is referred to as an excretory organ because it excretes (1) wastes. It is also a major homeostatic organ because it maintains the electrolyte, (2) and (3) balance of the blood. Urine is continuously formed by the (4) and is routed down the (5) by the mechanism of (6) to a storage organ called the (7). Eventually the urine is conducted to the body exterior by the (8). In males, this tube-like structure is about (9) inches long; in females, it is approximately (10) inches long.

Urinary System Coloring Workbook

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web nov 20 2020 in this video we ll balance the equation mg hno3 mg no3 2 no2 h2o and provide the correct coefficients for each compound to balance mg hno3 mg no3 2 no2 h2o you ll need to be

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web magnesium hydroxide nitric acid balanced total equation mg oh 2 aq 2 hno 3 aq 2 h 2 o mg no 3 2 aq total ionic equation mg 2 2 oh 1 2 h 1 2 no 3 1 2 h 2 o mg 2 2 no 3 1 net ionic equation 2 oh 1 2 h 1 2 h 2 o sulfuric acid and lithium hydroxide balanced total equation h 2 so 4 aq 2 lih aq 2 h 2