

1. Two groups of students (Group A and Group B) carried out a project\* on the chemistry of some group 7 elements (the halogens) and their compounds.

- (a) In the first part of the project, the two groups had a sample of iodine monochloride (a corrosive brown liquid) prepared for them by their teacher using the following reaction.



The following data were recorded.

|   |         |
|---|---------|
| Mass of $\text{I}_2(\text{s})$          | 10.00 g |
| Mass of $\text{Cl}_2(\text{g})$         | 2.24 g  |
| Mass of $\text{ICl}(\text{l})$ obtained | 8.60 g  |

- (i) State the number of significant figures for the masses of  $\text{I}_2(\text{s})$  and  $\text{ICl}(\text{l})$ . [1]

$\text{I}_2(\text{s})$ : .....

$\text{ICl}(\text{l})$ : .....

- (ii) The iodine used in the reaction was in excess. Determine the theoretical yield, in g, of  $\text{ICl}(\text{l})$ . [3]

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