

Chemical Reactions

📅 October 13, 2018

Physical vs chemical changes

It is important for you to understand the difference between **chemical** and **physical** changes. Some changes are obvious, but there are some basic ideas you should know.

Physical changes are usually about physical states of matter . Chemical changes happen on a **molecular** level when you have two or more molecules that interact. Chemical changes happen when **atomic bonds** are broken or created during chemical reactions .

Collision theory

At an atomic level, a chemical reaction will occur when two conditions are met. Two particles need to **collide** and they must have enough energy to react. This means the reaction rate will depend on these two factors: Collision rate & particle energy.

If two atoms collide but they don't have enough energy, then the reaction will not occur. If the particles have enough energy but they don't collide, then again, the reaction will not occur.

Rates And Reactions Study Guide

David R. Klein



Rates And Reactions Study Guide:

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