Chapter 6 Rates of Chemical Reactions

Practice Problems

Problem 1

The formation reaction of ammonia is represented by the following equation:

N_{high} + 3H_{high} → 2NH_{high}

Ammonia gas is produced at a rate of 4.0 × 10 3 mol/(L·s).

(a) What will be the rates at which the reactants are consumed?

(b) Assume that the reaction can proceed to completion. If the reaction begins with 1.0×10^2 mol of N_{2ij} and 2.0×10^2 mol of H_{2ij0} how much N_{2ij0} will remain when all H_{2ij0} is used up?

(c) If the reactants in (b) are placed in a 100 L reaction vessel and the reaction is allowed to proceed to completion, how long will it take?

Problem 2

For the reaction $5O_{2igh} + 4NH_{3igh} \rightarrow 6H_2O_{igh} + 4NO_{igh}$, the initial rate of production of H_2O is $2.64 \times 10^{-6} \text{ mol/}(L\text{-s})$. What is the corresponding rate of disappearance of O_2 ?

Problem 3

The rate law equation for the reaction $I_2 + (C_2H_3)_2(NH)_2 \rightarrow (C_2H_3)_2N_2 + 2HI$ is given below. Rate = $kI(C_3H_3)_4(NH)_2I$

The initial rate of reaction is 0.250 mol/L when the concentration of (C₂H₂)₂(NH)₂ is 30.0 mol/(L·s). What is the corresponding rate at which HI is formed if the initial concentration of (C₂H₂)₂(NH)₂ is 0.150 mol/L?

Problem 4

An organic reaction between complex molecules is represented by the following reaction.

 $2A + 2B \rightarrow C + 2D$

The rate of reaction is found to obey a rate law equation. Use the rate data below to answer the questions.

| Trial | Initial Concentration (mol/L) | | Initial Rate mol/(L-s) |
|-------|-------------------------------|-----|------------------------|
| | IAI | [8] | |
| 1 | 1.0 | 1.0 | 2.1×10^{-1} |
| 2 | 4.0 | 1.0 | 8.3 × 10 ⁻¹ |
| 3 | 1.0 | 2.0 | 8.4 × 10 ⁻¹ |

(a) What is the overall reaction order of this reaction?

(b) Calculate the value of the rate constant.

(c) What will be the effect upon the rate of reaction if the concentrations of both A and B are doubled?

Rates Of Reaction 18 Practice Problems Worksheey

Karin Nielsen-Saines

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Introduction to Chemistry, Laboratory Manual T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics. The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use **Holt Chemistry** Ralph Thomas Myers,2004 Teacher's Wraparound Edition: Twe Biology Everyday to students **Experience** Albert Kaskel, 1994-04-19 Active Living Every Day Steven N. Blair, Andrea L. Dunn, Bess H. Marcus, Ruth Ann Carpenter, Peter Jaret, 2021 Active Living Every Day Third Edition is a behavior change strategy book intended to help the reader become more physically active It includes activities and tools within every chapter to encourage the reader to make long term changes to improve health CBSE Chapterwise Worksheets for Class 9 Gurukul, 2021-07-30 Practice Perfectly and Enhance Your CBSE Class 9th preparation with Gurukul s CBSE Chapterwise Worksheets for 2022 Examinations Our Practicebook is categorized chapterwise topicwise to provide you in depth knowledge of different concept topics and questions based on their weightage to help you perform better in the 2022 Examinations How can you Benefit from CBSE Chapterwise Worksheets for 9th Class 1 Strictly Based on the Latest Syllabus issued by CBSE 2 Includes Checkpoints basically Benchmarks for better Self Evaluation for every chapter 3 Major Subjects covered such as Science Mathematics Social Science 4 Extensive Practice with Assertion Reason Case Based MCQs Source Based Questions 5 Comprehensive Coverage of the Entire Syllabus by Experts Our Chapterwise Worksheets include Mark Yourself at the end of each worksheet where students can check their own score and provide feedback for the same Also consists of numerous tips and tools to improve problem solving techniques for any exam paper Our book can also help in providing a comprehensive overview of important topics in each subject making it easier for students to solve for the exams Chemical Process Engineering, Volume 2 A. Kayode Coker, Rahmat Sotudeh-Gharebagh, 2022-07-20 CHEMICAL PROCESS ENGINEERING Written by one of the most prolific and respected chemical engineers in the world and his co author also a well known and respected engineer this two volume set is the new standard in the industry offering engineers and students alike the most up do date comprehensive and state of the art coverage of processes and best practices in the field today This new two volume set explores and describes integrating new tools for engineering education and practice for better utilization of the existing knowledge on process design Useful not only for students university professors and practitioners especially process chemical

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