

## SALT ANALYSIS – SALT 4

**AIM OF THE EXPERIMENT :** TO ANALYSE THE GIVEN SALT FOR ACIDIC AND BASIC RADICALS .

EXPERIMENT	OBSERVATION	INFERENCE
<b><u>PHYSICAL EXAMINATION OF THE SALT</u></b> 1. COLOUR  2. ODOUR  3. FLAME TEST A small amount of salt paste with conc. HCl is place on platinum wire loop and introduce into the flame .	White  Ammonical and vinegar like smell  No characteristic observation	$\text{Cu}^{2+}$ , $\text{Mn}^{2+}$ , $\text{Co}^{2+}$ , $\text{Ni}^{2+}$ , - $\text{Fe}^{3+}$ are absent $\text{NH}_4^+$ or $\text{CH}_3\text{COO}^-$ may be present .  $\text{Ca}^{2+}$ , $\text{Ba}^{2+}$ $\text{Sr}^{2+}$ , $\text{Na}^+$ , $\text{K}^+$ , $\text{Cu}^{2+}$ , $\text{Zn}^{2+}$ , $\text{Mn}^{2+}$ , $\text{Pb}^{2+}$ are absent
<b><u>FOR ACID RADICALS</u></b> <b><u>(i) PRELIMINARY TEST WITH DILUTE <math>\text{H}_2\text{SO}_4</math></u></b>  Salt sample + 1- 2 ml dil. $\text{H}_2\text{SO}_4$ . Noted the changes in cold condition and then the solution is gently made warm.	No characteristics observation ie no gas evolved	$\text{CO}_3^{2-}$ , $\text{NO}_2^-$ , $\text{S}^{2-}$ , $\text{SO}_3^{2-}$ etc are absent
<b><u>(ii) PRELIMINARY TEST WITH CONC. <math>\text{H}_2\text{SO}_4</math></u></b> Salt sample + about 1- 2 ml. of conc. $\text{H}_2\text{SO}_4$ and warmed gently.	A colourless vapours evolved	May be $\text{Cl}^-$ , $\text{Br}^-$ , $\text{I}^-$ . $\text{NO}_3^-$ , $\text{CH}_3\text{COO}^-$ , $\text{C}_2\text{O}_4^{2-}$ are present
<b><u>CONFIRMATORY TEST FOR <math>\text{NO}_3^-</math></u></b> <b><u>1. COPPER CHIPS TEST</u></b> Salt sample + Conc. Sulphuric acid + few copper chips  <b><u>2. RING TEST</u></b> Salt sample + freshly prepared ferrous sulphate solution + conc . sulphuric acid added slowly along the sides of the test tube .	Dark brown fumes of nitrogen dioxide are evolved .  A dark brown ring is formed at the junction of the layers of the acid and the solution	Presence of $\text{NO}_3^-$ is confirmed .  Presence of $\text{NO}_3^-$ is confirmed .

**FOR BASIC RADICAL**

# Salt Analysis Manual For P G Course

**Guillaume Favre**



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