

Elements of the Sea

Ionic equations

- If 2 solutions react to form a solid, a precipitation reaction occurs. Eg. $\text{Ag}^+(\text{aq}) + \text{NO}_3^-(\text{aq}) + \text{Na}^+(\text{aq}) + \text{Br}^-(\text{aq})$ goes to $\text{AgBr}(\text{s}) + \text{Na}^+(\text{aq}) + \text{NO}_3^-(\text{aq})$

Then cancel spectators

$\text{Ag}^+(\text{aq}) + \text{Br}^-(\text{aq})$ goes to $\text{AgBr}(\text{s})$

- Always include state symbols
- Only split into ions if **aqueous and ionic**

Ionic solids

- Cations and anions are held together by strong electrostatic attractions in a **giant ionic lattice**.
- Each sodium ion is surrounded by 6 chloride ions **6:6 coordination**. It is a **simple cubic** structure.
- Ionic solids are hard and have a high melting point due to strong electrostatic attractions between ions.
- If ionic crystals contain water, the water molecules fit into the lattice the same way as regular ions. This is called the **water of crystallisation**. These crystals are said to be hydrated.
- If these hydrated crystals are heated, the water molecules are driven off and you are left with an **anhydrous solid**.



Ionic substances in solution

- As ionic substances dissolve in water, the ions become surrounded by water molecules and they spread throughout the solution. The hydrated ions are **randomly** arranged and behave **independently**.

Salters Elements Of Sea Module Test

Stephen John Salter



Salter's Elements Of Sea Module Test:

Salter's Higher Chemistry, 1999 This work provides coverage of the content statements in the arrangements for Higher Chemistry organized by the three units in the course Energy Matters the World of Carbon and Chemical Reactions At the start of each unit students are given guidance on what they need to know and understand *The Marine Environment* J. M. A. Lenihan, William Whigham Fletcher, 1977 **The Chemical Engineer**, 1976 **Solar Energy Update**, 1978-04 **Chemical Engineering Catalog**, 1925 **The Engineer**, 1915 **SBASI** Stephen John Salter, 1975 **Scientific and Technical Aerospace Reports**, 1977 *SBASI* S. J. Salter, Raymond Eugene Lundberg, G. L. McDougal, 1975 **The Electrical Review**, 1906 Engineering, 1923 The Independent Leonard Bacon, Joseph Parrish Thompson, Richard Salter Storrs, Joshua Leavitt, Henry Ward Beecher, Theodore Tilton, Henry Chandler Bowen, William Hayes Ward, Hamilton Holt, Fabian Franklin, Harold de Wolf Fuller, Christian Archibald Herter, 1928 Nuclear Science Abstracts, 1971-11 *The Builder*, 1881 **Harper's Weekly**, 1890 **Energy Research Abstracts**, 1988 *Machinery Lloyd*, 1953 Fuel and Energy John Hadlett Harker, J. R. Backhurst, 1981 **Amateur Gardening**, 1967 **World Fisheries** Alexander Fridman, 1998

Salters Elements Of Sea Module Test Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Salters Elements Of Sea Module Test**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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