- a) What are Carbenes? What is the state of their hybridisation? Discuss structure and stability of singlet and triplet dichlorocarbene.
- b) Briefly explain the following observations: 2 × 2
 - i) Phenyl anion is more stable than cyclohexyl anion.
 - ii) N, N-dimethylaniline is weak as a base thatnN, N, Z, 6 tetramethylaniline.
- c) Methane reacts with Cl₂ to form CH₃Cl and HCl, but with I₂ doesn't occur at appreciable rate, even I I bond is much weaker than Cl Cl bond. Explain why this is SO.
- d) What products are formed when glycine (H₂NCH₂CO₂H) is treated with conc. HCl and NaOH separately.
- 2. a) Draw the structure for each of the following compounds:
 - i) (1R, 3S) 1, 3-cyclohexanediol
 - ii) (1Z, 3E)-1-bromo-1, 3-pentadiene.

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RANJAN KUMAR MOHAPATRA

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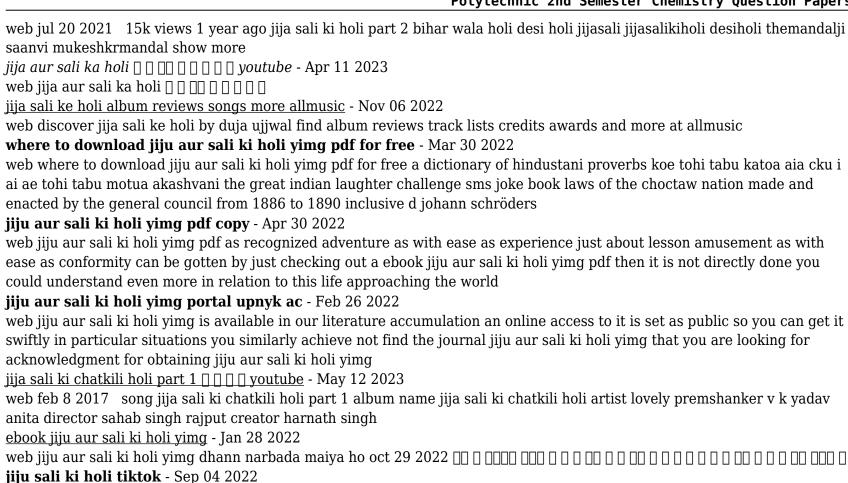
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