MKC STUDY GUIDE

REVERSE OSMOSIS (MK3)

- Theory: Pressurized seawater is passed over a semi permeable membrane that passes pure water but excludes salt water.
- Osmotic pressure is the difference in pressure across the membrane.
- Reverse Osmosis is known as Hyper-filtration.
- Osmotic pressure is equal to 1 psi per 100 milligrams per liter of salt.
- The osmotic pressure of sea water is 350-370 psi.
- Osmotic pressure occurs when pressure is applied to the higher salt concentration side of the membrane, causing pure water to be forced out of the salty solution.
- Low Pressure Switch: Secures the high press pump if the low press switch predetermined setting isn't met (on 110's setting is 5 psi).
- High Pressure Pump: Creates a pressure of 700-1000 psi to achieved reverse osmosis.
- Pulsation Dampener: Absorbs pulsations created on discharge side of high press pump. Consists of a nitrogen charged rubber bladder charged to approximately 2/3 system pressure.
- Cracking pressure for system relief valve is most likely set at 1050 psi.
- Pressure Regulator: Normally a needle valve installed on the brine discharge line used to keep the module pressure at 700–1000 psi.
- When the salinity cell detects more than 500 ppm tds (salt) it shifts the 3 way valve to overboard discharge position. Audio & visual alarm notification.
- Crossflow: Allows the membrane to continually clean itself. Fluid passes through the membrane while the rest continues downstream, sweeping the rejected species away from the membrane.
- Daltons: Atomic Mass Unit (measure of atom's mass). A unit of mass defined as 1/12 of the mass of a carbon-12 atom, equal to 1.66 x 10
 kg.
- John Dalton: British physicist and chemist that developed the atomic theory of elements and molecules, founder of modern physical science.

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