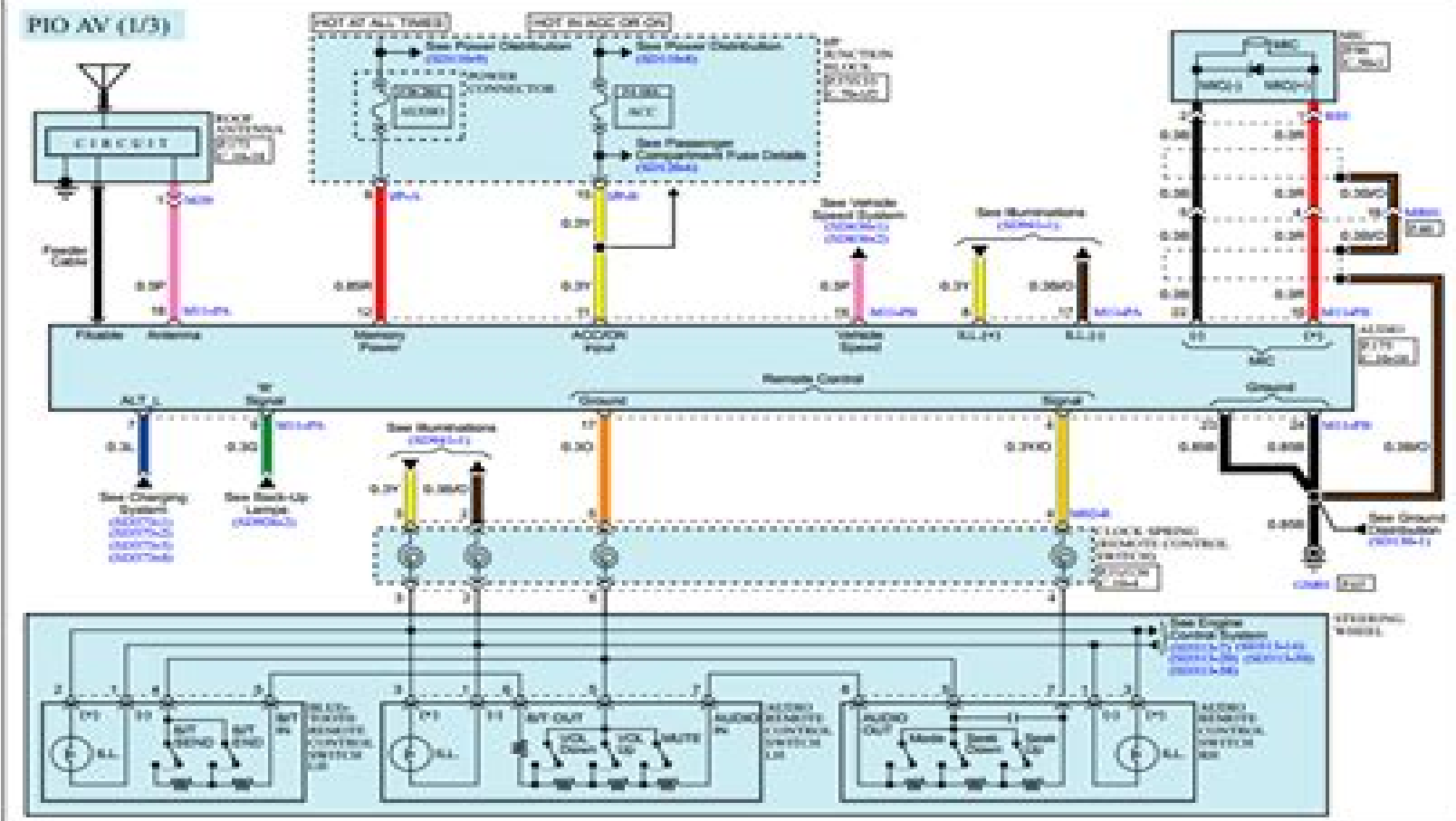


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web record the atomic mass and abundance in nature of certain isotopes explain how abundance in nature influences average atomic mass note this activity requires access to the phet interactive check it out here isotopes atomic mass purchase includes the 3 page student worksheet pdf 4 page teacher answer key pdf

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web multiple choice 1 minute 1 pt lithium has only two stable isotopes use the sim to determine the average atomic mass of a sample containing three lithium 6 atoms and two lithium 7 atoms 6 51556 amu 6 61565 amu 6 34875 amu 6 41548 amu

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web write at least three conclusions from your exploration 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3 the more unstable the atom the less presence or no presence in nature 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3

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web isotopes and atomic mass model 1 make isotopes open the isotopes and atomic mass simulation play with the make isotopes tab of the simulation for a few minutes and then answer the following questions 1 what particles determine the mass number number of protons and neutrons 2 why is mass number always a whole

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web this guided inquiry activity has two learning goals students will be able to students will be able to 1 explain the difference between atomic mass and mass number and 2 calculate average atomic mass from percent abundance and isotopic mass

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web find the average atomic mass of an element given the abundance and mass of its isotopes predict how the mass and name of an isotope will change given a change in the number of protons neutrons or electrons

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