

- 4 Hydrogen and chlorine are reacted together to form hydrogen chloride.



- (a) Calculate the bond enthalpy for the H–Cl bond using the information in the table below.

bond	bond enthalpy / kJ mol^{-1}
H–H	+436
Cl–Cl	+243

Bonds broken

436

- 243

= 679

-679

Bonds made 2(HCl)

Bonds Broken + Bonds Made = -184

-679 + 2HCl = -184

2HCl = 495

HCl = 247.5

bond enthalpy = kJ mol^{-1} [2]

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LP Steffe

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'LP Steffe' text.

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